

The Russian R&D in the field of Hydrogen Technologies

A faint, light blue map of Russia is visible in the background of the slide, showing the country's extensive territory across Eurasia.

S.P. Malysenko

Laboratory for Hydrogen Energy Technologies

Joint Institute for High Temperatures

(IVTAN H2Lab)

Predicted hydrogen production and prices on Russian market

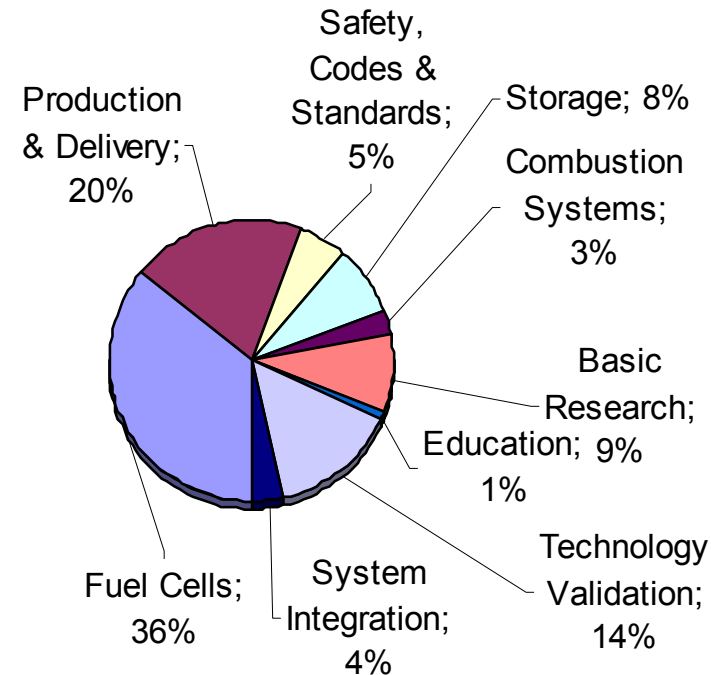
	2025	2035	2045	2055
Production, 10⁶ toe				
European part	2-3	15-50	20-30	15-25
Ural and Western Siberia	14-17	15-40	20-25	15-20
Eastern Siberia and Far East	5-6	10-15	15-20	20-25
Total	21-26	40-105	55-75	50-70
Prices, \$/toe				
European part	245-250	260-270	350-370	470-490
Ural and Western Siberia	215-220	245-250	295-310	470-490
Eastern Siberia and Far East	165-175	185-195	200-215	240-260

2006, Nekrasov A.S., Sinyak Yu.V.
RAS

Russian Hydrogen Program: 2005-2006 Federal Budget

- Ministry of Education and Science, Federal Agency for Science and Innovations (FASI)
- Russian Academy of Sciences
- Russian Foundation for Basic Research
- Federal Agency for Nuclear Energy (Minatom)
- Federal Space Agency

- Federal Program Research and Innovations in Priority Trends of Scientific and Technical Development for 2002-2006
- Educational Programs
- Basic Research Projects
- Technology Oriented Projects



Total: ~\$15 M annual

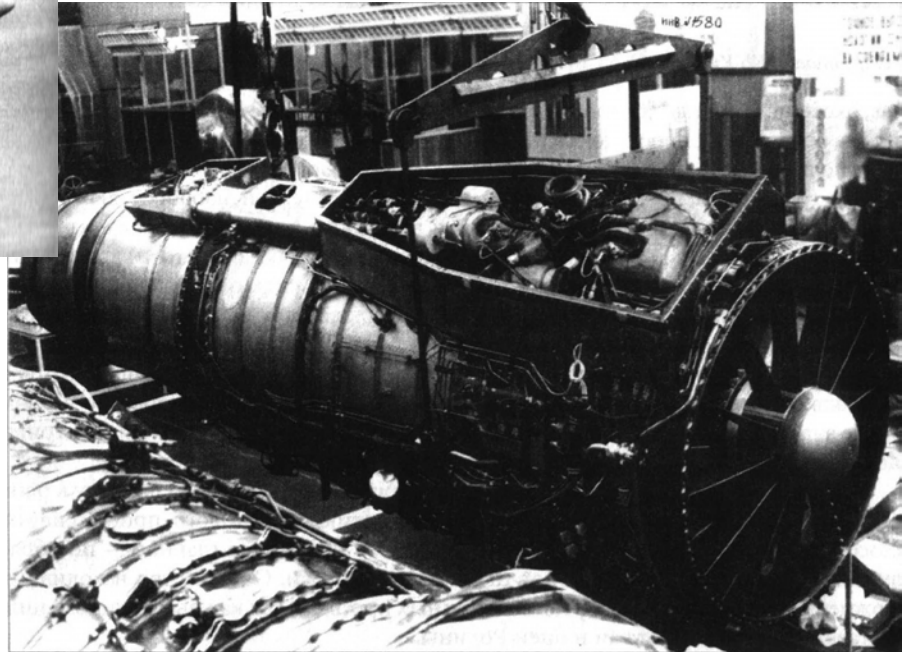
Additional funding:
private companies (Norilskii Nickel, Gazprom, etc.), Moscow government

Aerospace Hydrogen Technologies Experience



Hydrogen engine NK-88

1988. First hydrogen flight. Tu-155.



Hydrogen for space program

1972

Automatic flight to the Moon



- **HYDROGEN – 20 KG**
 - **OXYGEN – 200 KG**
- FUEL CELL POWER UNIT***

1987

Energia



HYDROGEN – 106 TONNS
OXYGEN – 1540 TONNS
ROCKET ENGINES

1988

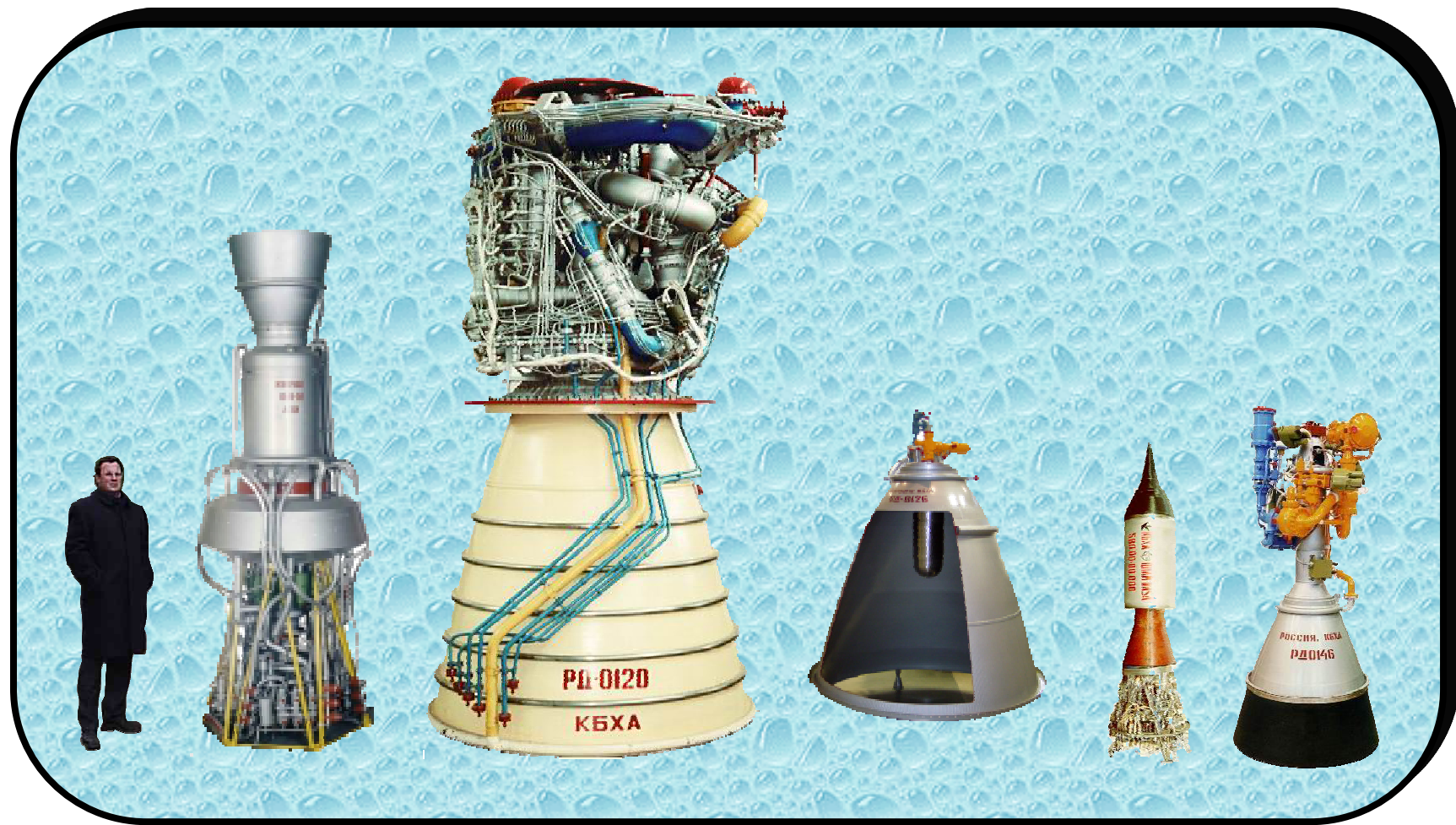
Energia-Buran



- **HYDROGEN – 110 KG**
 - **OXYGEN – 900 KG**
- FUEL CELL POWER UNIT***

Hydrogen engines

Chemical Automatics Design Bureau, Russian Space Agency



Modernization of space launch vehicles

SPACE LAUNCH VEHICLES

WEIGHT, T

USEFULL LOAD, T:

- circular orbit=200 KM
- GEO
- GTO ($\Delta V=1500$ M/S)

ENGINES

- I STAGE
- II STAGE
- III STAGE

FUEL

«SOUZ-FG»

BAIKONUR

308,0

7,44

-

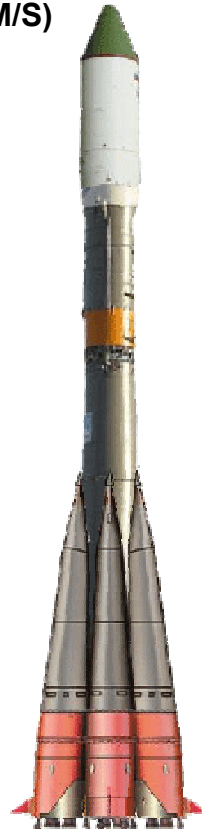
1,5

14Д22

14Д21

11Д55

OXYGEN +
KEROSINE



«ONEGA»

BAIKONUR

391

15,6

2,7

4,8

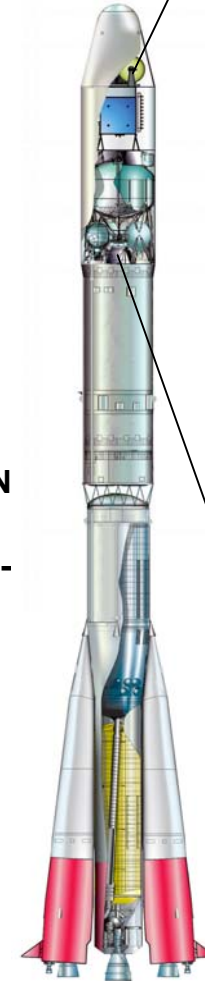
РД-120.10Ф

РД-191

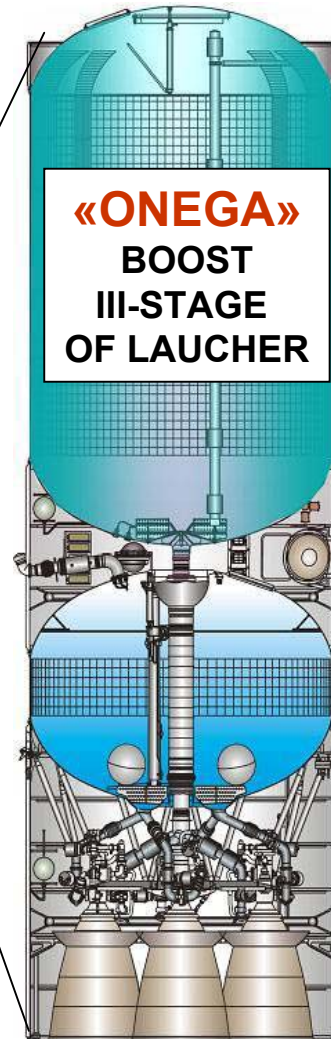
РД 0146Э

I, II STAGES – OXYGEN
+ KEROSINE

III STAGE AND BOOST -
**OXYGEN +
HYDROGEN**



«ONEGA»
BOOST
III-STAGE
OF LAUCHER





Federal Agency for Science and Innovations

- Program: Research and Innovations in Priority Trends of Scientific and Technical Development for 2002-2006
- Priority direction: Energy and Energy Saving

**Total: \$15 M for
R&D in Hydrogen Technologies
in 2005-2006**



FASI Hydrogen Program 2005-2006 Projects

- Modular Power Units with 10 kW SPE FC
- Power Production System with 5 kW SO FC
- Portable and micro FC
- Portable FC Processors for Hydrogen Power Supply Systems with FC
- Reversible Solid State Hydrogen Storage and Purification Systems for FC
- Reversible Solid Oxide Electrochemical Devices (electrolyzer – FC system)
- Non-Pt Catalysts for FC
- H₂/O₂ Steam Generators for Steam-Turbine Power Units
- Autonomous Power Supply Systems on the Base of Renewable Energy Sources with Hydrogen Energy Accumulation
- Safety and Standards for Hydrogen Energy
- Education

International Co-operation

- ☞ IPHE Collaborative Projects
- ☞ The International Science and Technology Center (ISTC), ISTC Fuel Cell Initiative
- ☞ US Civilian Research and Development Foundation
- ☞ EU Framework Program (FP6)
- ☞ by-lateral cooperation

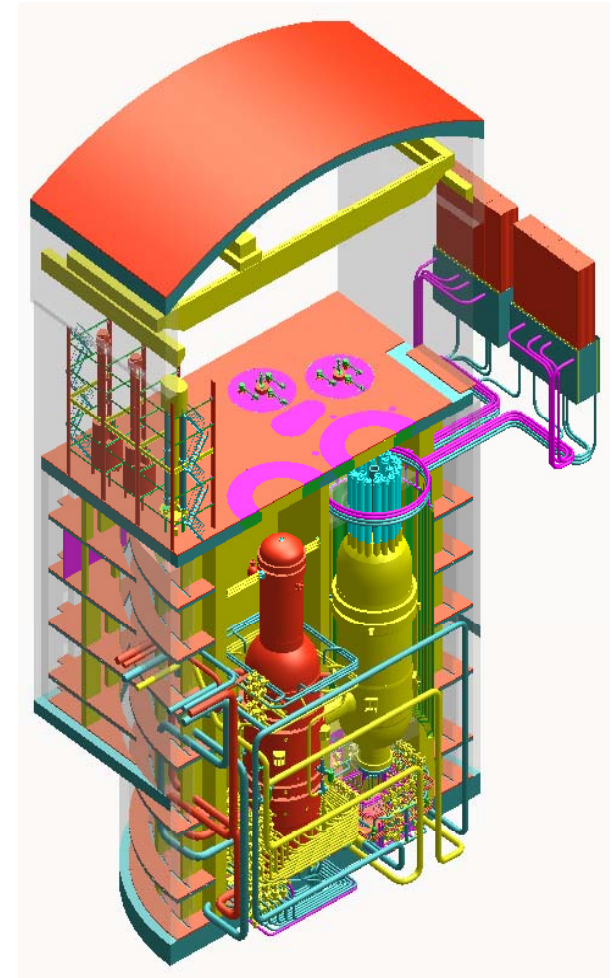
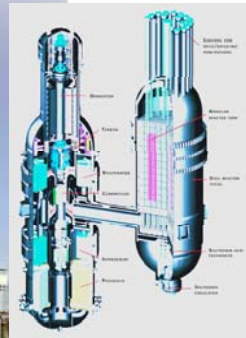
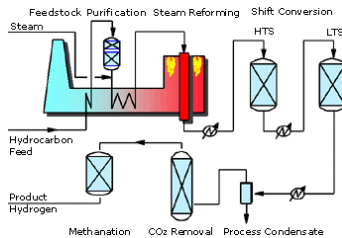
NEW HYDROGEN PRODUCTION TECHNOLOGIES

INNOVATIVE NATURAL GAS REFORMING FOR HYDROGEN PRODUCTION



Hydrogen Energy & Plasma Technology Institute

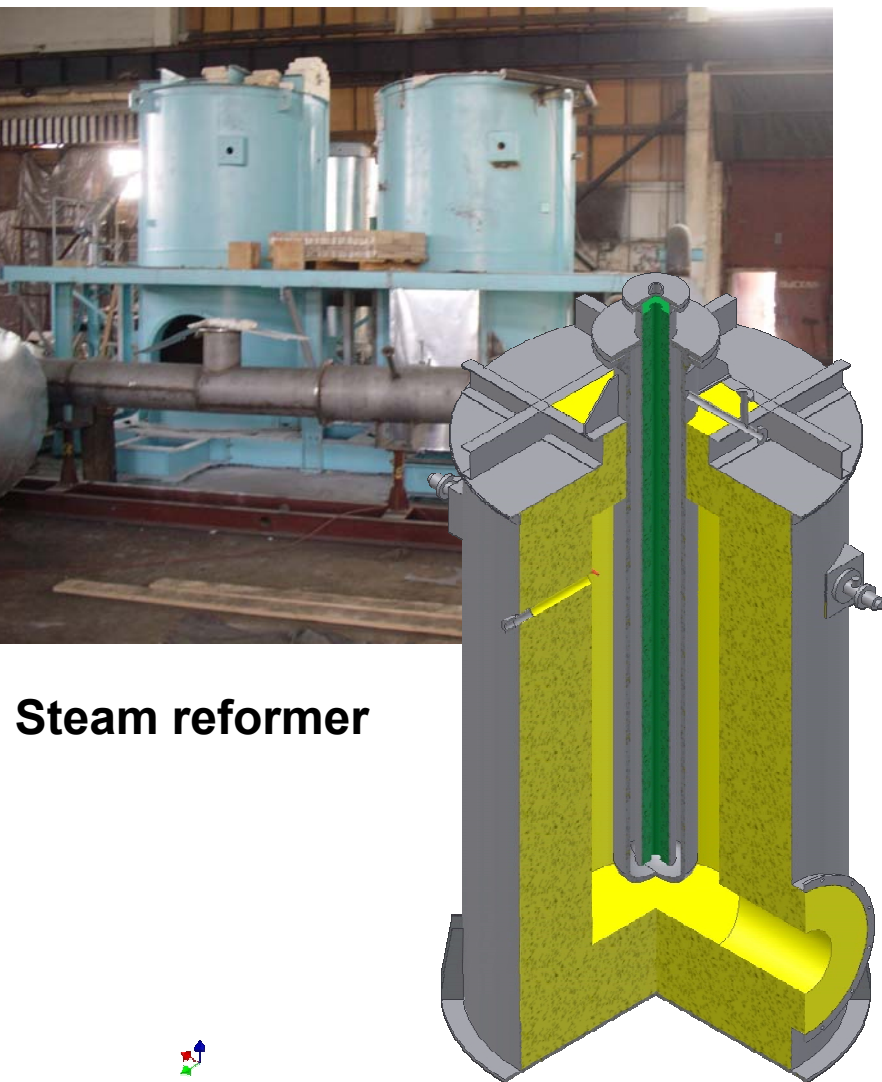
Steam conversion of methane in MGR-T



- H_2 yield two times higher than for pyrolysis of methane

Hydrogen production

Mashprom



Productivity	250 s m ³ /h
Purity	99,98% *
Hydrogen pressure in receiver	0,5-1,0 MPa
Energy consumption	60 kW h
CO ₂ production	150 kg/h
Dimensions, m:	
conversion block	12 x 8 x 4,5
purification block	8,5 x 8 x 4,5
PSA purification	10 x 8 x 6
Weight, t :	
conversion block	60
purification block	27
PSA purification	20

Steam reformer



NATURAL GAS PROCESSING

FOR PRODUCTION OF HYDROGEN AND PURE CARBON MATERIALS

Institute for High Temperatures RAS

On the basis of two-stage pyrolysis of natural gas the technology to produce hydrogen from natural gas simultaneously with pure carbon material for broad commercial applications has been developed.

As a porous carbon skeleton for pyrocarbon stuffing other carbon containing materials can be used (e.g. wood waste products, generally the vegetable origin waste).

Address: IVTAN, 125412, Izhorskaya 13/19, Moscow

E-mail: zaitch@oivtran.iitp.ru



0.01	Moisture content, %	0.52
0.43	Volatile, %	1.12
0.04	Ash, %	1.54
0.02	Sulphite, %	0.30
99.56	C, %	97.38
0.25	H, %	0.27
33.18	Calorific value, MJ/kg	32.57
1.62	Density, g/cm ³	0.742

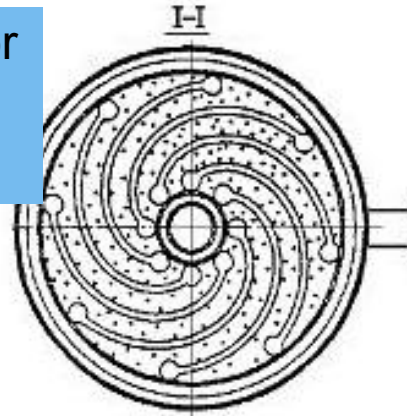
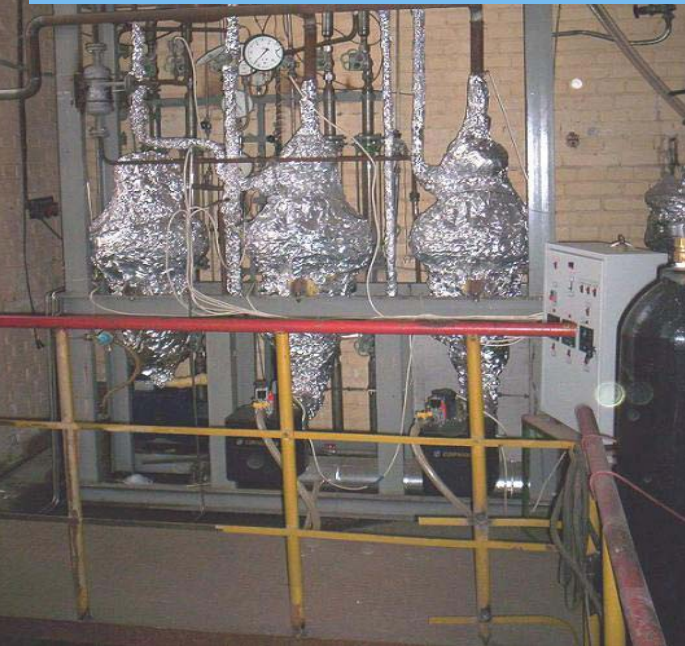


COMPACT HIGH EFFECTIVE HYDROGEN PRODUCTION BY STEAM NATURAL GAS CONVERSION

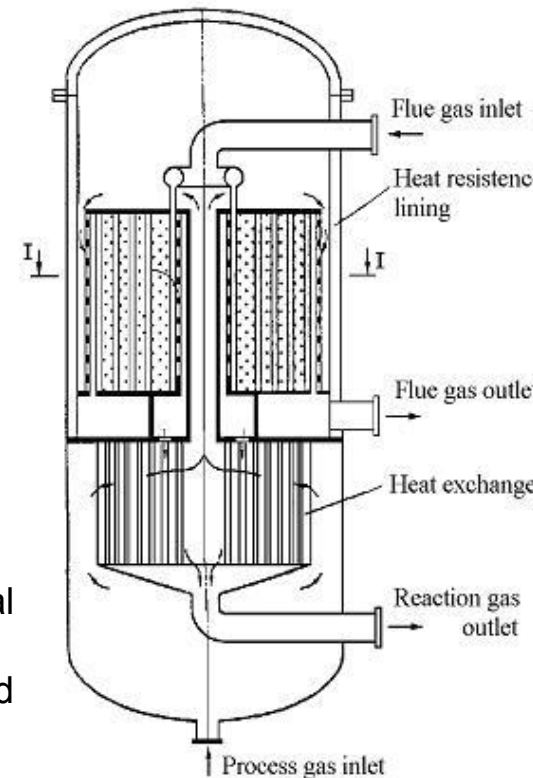
FAST ENGINEERING Ltd.

Pilot synthesis gas production unit for under 6,4 MPa.

FAST ENGINEERING® PROCESS



- No oxygen using
- The using the most active fine-grained catalyst for steam natural gas conversion keeping low pressure drop in the granular bed and allowing to increase a few time volume wear rate

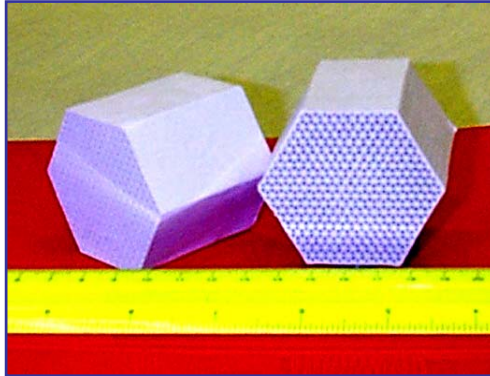


- Required heat supply for carrying out endothermic reaction of steam conversion
- Reducing of operation costs
- An opportunity to create steam conversion for required capacity
- Reducing pollution to environmental
- Compactness

«FAST ENGINEERING», Moscow, 117218, RUSSIA
Tel./Fax: +7 495 718-8196, astanovsky@fastmail.ru;
<http://www.fastmail.ru/~astanovsky>

Catalysts and reactors for partial oxidation of natural gas into a mixture of H_2 and CO

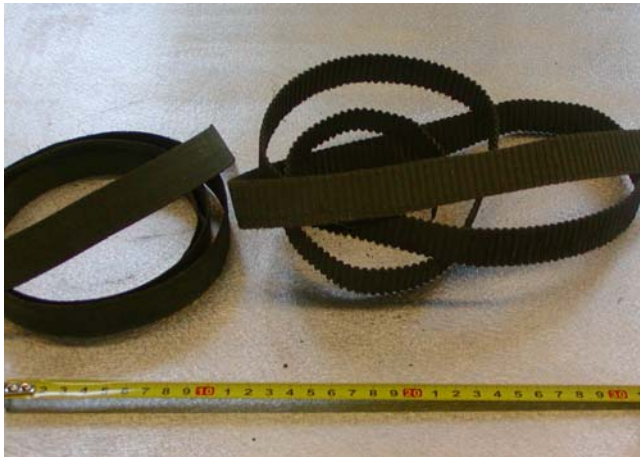
Research Institute of Catalysis, Novosibirsk, Russia



Monolithic catalysts



Monolithic catalyst reactor



Tape catalyst



***Tape catalyst reactor
of the syngas capacity ca. 5 m³/hour***

prototype of the fuel processor based on an integrated reformer unit for steam reforming of natural gas in a 5 kW_e SOFC plant



Fuel processor

The fuel processor includes:

- A unit for desulfurization of natural gas
- An integrated reforming unit
- A system for feeding natural gas, water, air
- Control sensors for temperature, pressure, reagent consumption
- An automated control system for the fuel processor (monitoring of temperature, pressure, consumption, as well as control of reagent consumption, stop and switch valves, a starting device)

Test data for the fuel processor:

- Stable operation on varying the natural gas consumption from 0.1 to 0.9 m³/h
- Capacity up to 5 m³/h for synthesis gas
- Hydrogen content up to 75 vol % in synthesis gas

Microcatalytic hydrogen production systems and reactor for steam conversion of methanol



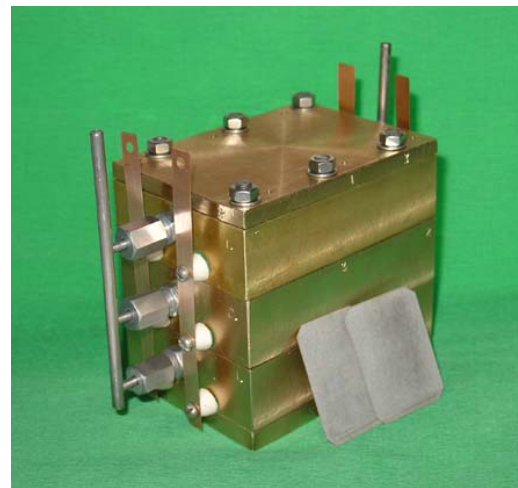
Boreskov Institute of Catalysis, Novosibirsk, Russia



Microreactor with external heating



Microreactor with heat exchanger



Sectioned high power reactor

Characteristics

Dimensions		
Ø30 h=60mm	Ø40 h=35 mm	65×90×85 mm
Microchannel plate		
Ø28 δ=0,3 mm	Ø35 δ=0,3 mm	30×40 mm
Hydrogen output and heat power		
130 l/h; 300 W	32 l/h; 73 W	320 l/h; 750 W

ALKALI ELECTROLYSIS

Moscow Power Engineering Institute (TU)
Joint-Stock Company "UralChemMash"

New alkaline electrolysis cell

High effective electrodes without precious metals content

Power consumption 4,2 kW-hour/s M³
under current density 3 kA/M²

Purity of hydrogen at battery
output up to 99,7 %

Absence of asbestos diaphragm

Compactness

Possibility of hydrogen consumption
regulation at wide range

New battery for alkali electrolysis

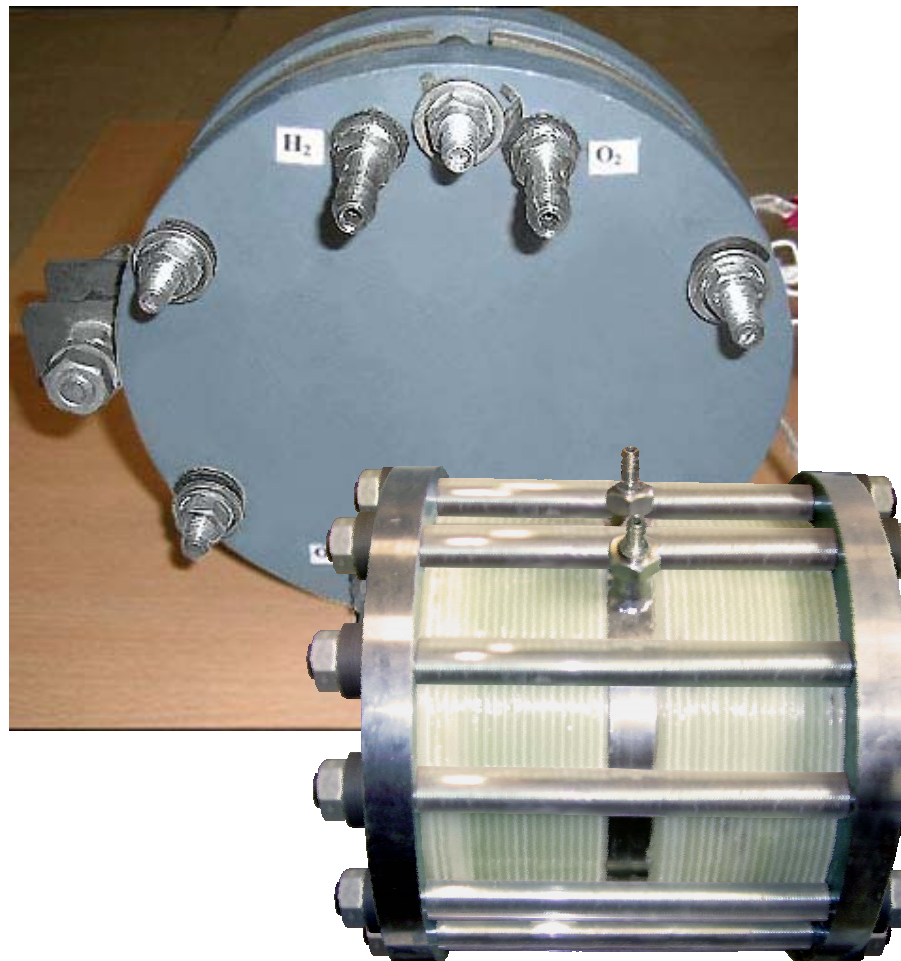
Productivity – 1 s M³ H₂/hour

Power consumption 4,2 kW-hour/s M³
under current density 3 kA/M²

Purity of hydrogen at battery
output up to 99,7 %

Compactness

Possibility of hydrogen consumption
regulation at wide range



PEM Electrolyzers for high purity hydrogen production

At present time the fields of applications of PEM-electrolyzers are: fuel cells, gas chromatography, water-chemical regime correction systems of nuclear reactors, hydrogen welding, metallurgy, electronic industry, analytical chemistry, etc.

PEM-electrolyzer and plants on its base with various productivity (from several ml to tens cubic meters per hour) and purposes have been developed

Technical performances of PEM electrolyzers:

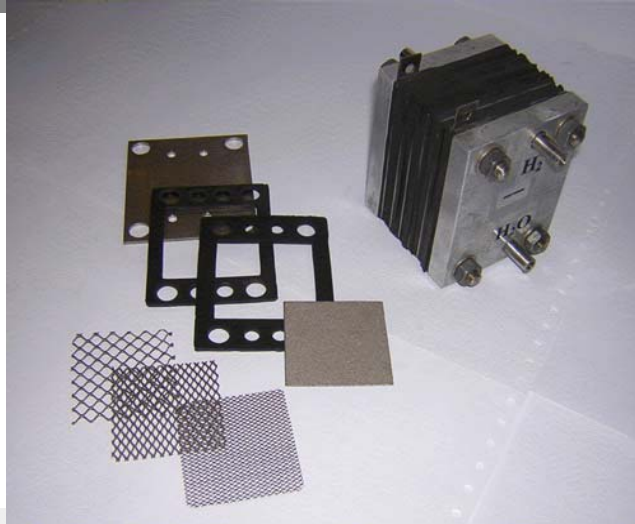
Power consumption 3.9-4.1 kW·hour/1m³ of H₂

Voltage on the cell 1.67-1.72 V at current density 1 A/cm² and t=90°C

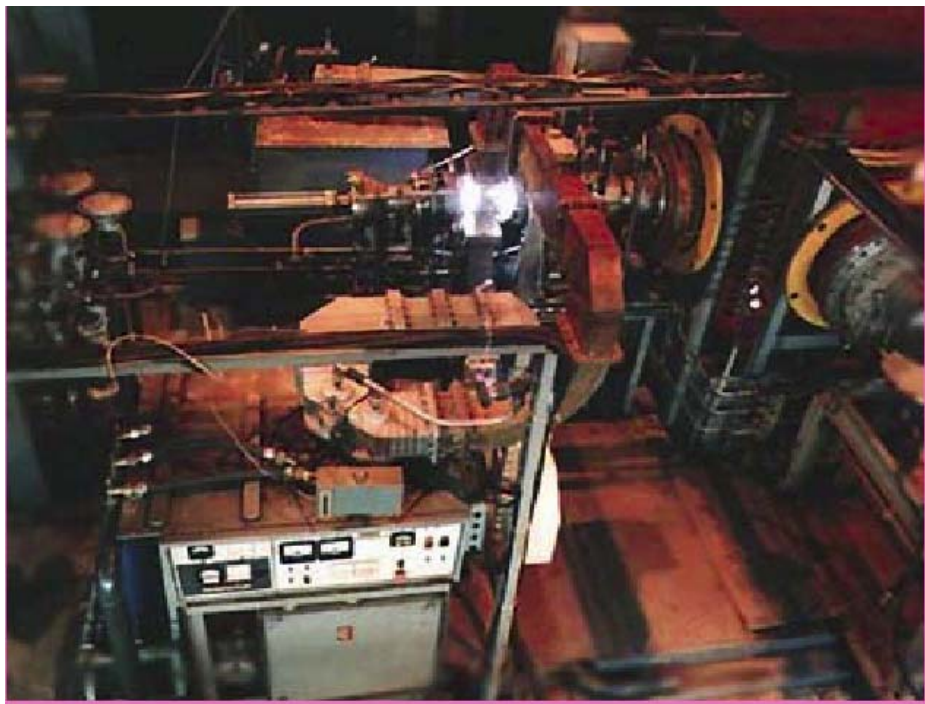
Operating pressure up to 3,0 MPa

Hydrogen purity > 99.99%

Life time (average life) > 5000 hours



RRC “Kurchatov Institute’s” Test Facility “POVOD” for Development and Scaling of Plasma Assisted Processes of Hydrogen and Syn-Gas Production



“POVOD” demonstration unit at RRC “Kurchatov Institute” was designed to investigate and demonstrate the whole set of gas-phase plasma chemical processes under effect of stationary microwave discharge with power range from 10kW to 1,000 kW at microwave frequency 915 MHz. Air, nitrogen, water vapor, carbon dioxide, methane, propane-butane, oxygen, argon with a flow rate up to 2,000 m³/h at pressure between 0.005 atm – 1.0 atm can be used to burn MW discharge.

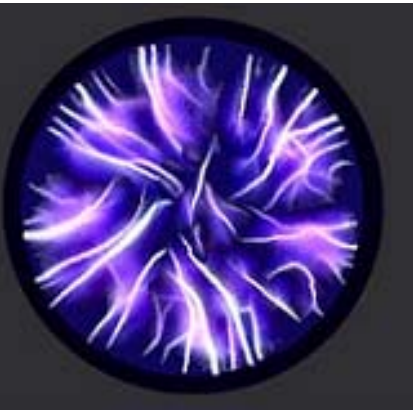
Address: HEPTI RRC “Kurchatov Institute”, 123182, Moscow, Russia

Tel: (7-495)1969439; **Fax:** (7-495)1966278

E-mail: s.korobtsev@hepti.kiae.ru

Microwave reformers of hydrogen raw material conversion into synthesis gas

Two converter types have been developed: based on the impulse periodical microwave discharge, and on the stationary discharge.



Stationary converter

Synthesis gas production: 20 nm³/hour;

Converter size: below 350 x 500 x 500 mm,

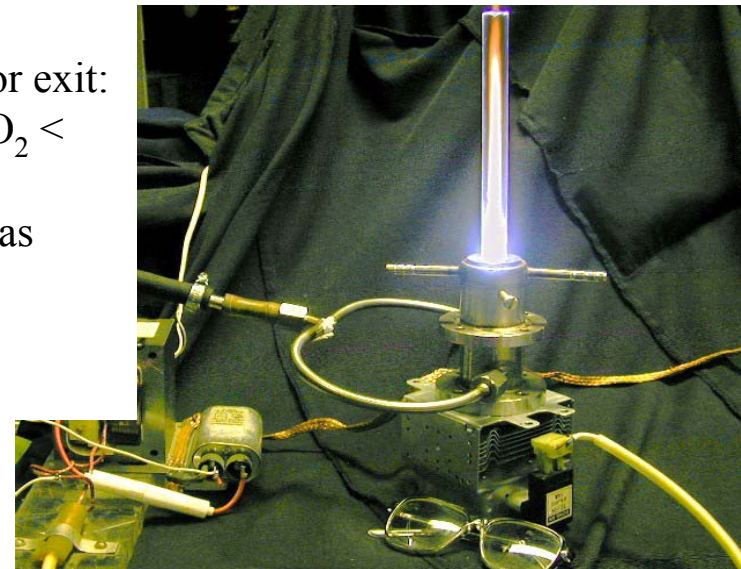
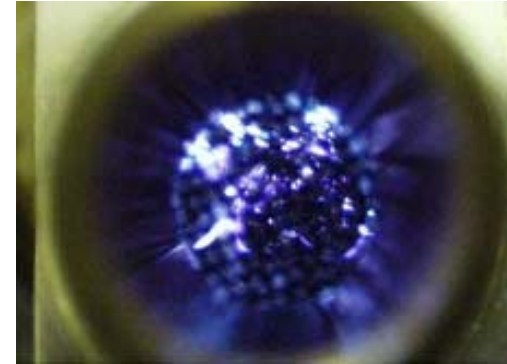
Mass: 31.5 kg, converter volume: 0.1 m³.

Typical gas composition at reactor exit:

N₂ < 54.2%; Syngas > 40.0%; CO₂ < 2.5%; CH₄ + C₂H₄ + C₂H₂ < 3.3%

Electric power expenses for syngas production 0.15 kWh/m³

Conversion efficiency (with heat recuperation) is 94%.



123182, Moscow, Russia

Tel: (7-495)1969439; Fax: (7-495)1966278

s.korobtsev@hepti.kiae.ru

HYDROGEN GENERATION FROM LOW-COST ALUMINUM



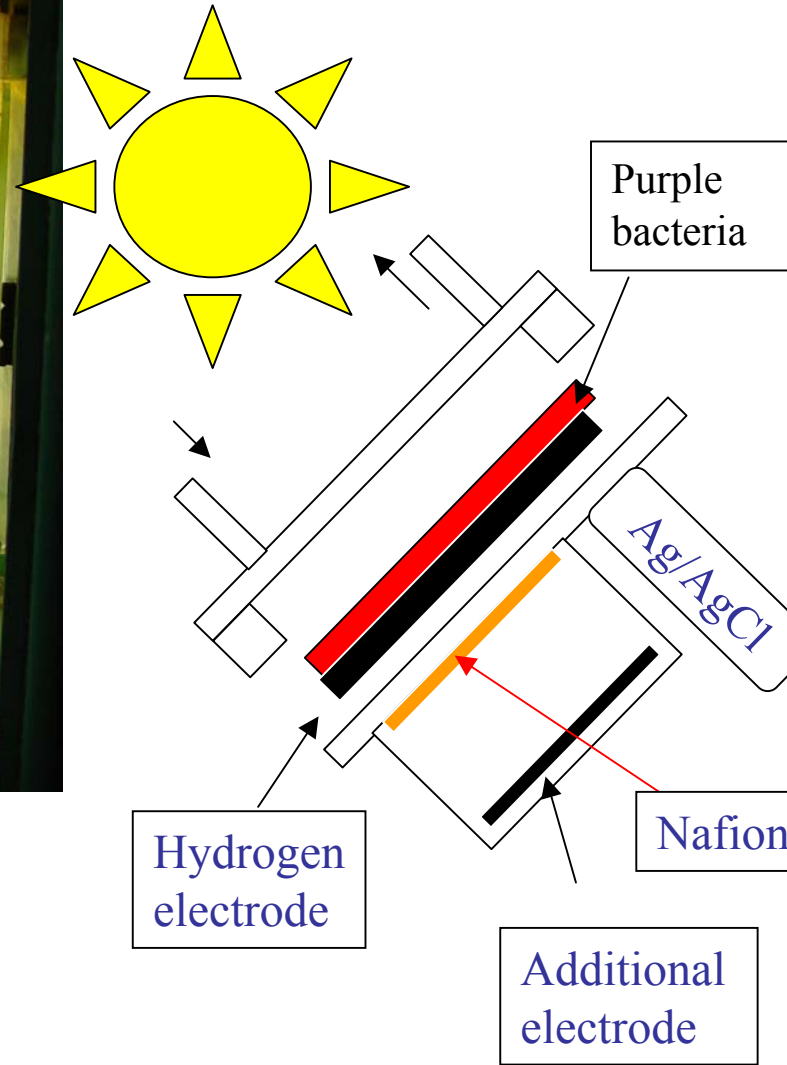
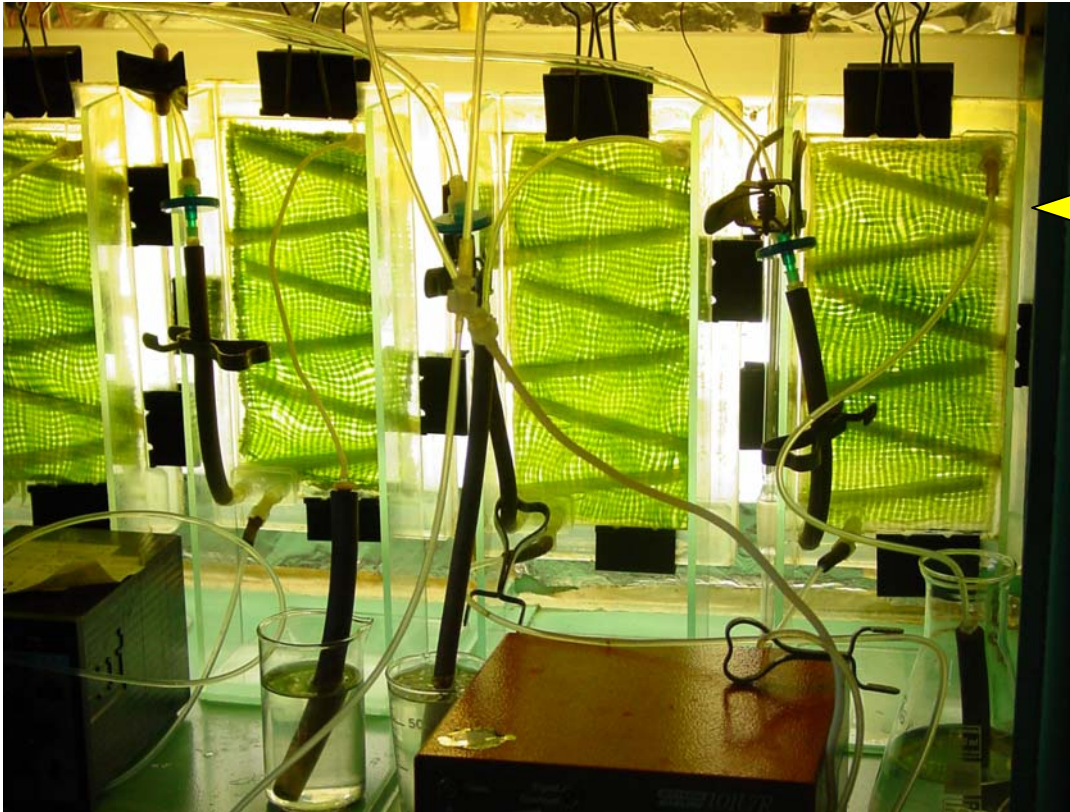
System for high temperature (up to 1000° C) Hydrogen generation.

4 m³/h of H₂ (0.363 kg/h) and 6 kg/h of nanocrystalline Al – hydroxide.

Science and Engineering Center for Energy Saving Processes and Equipment. Russian Academy of Science (SEC ESPE)

BIOCONVERSION OF REVERSIBLE ENERGY SOURCES

Institute of Fundamental Problems of Biology, RAS



FUEL CELLS

International Project on Fuel Cells Based on Nanoporous Structures

Association for Advanced Technologies `ASPECT`

Institute of Physical Chemistry and
Electro-Chemistry after Frumkin
(IPCEC)

Institute for High Temperature (IHT)

Institute of Petrol-Chemical Synthesis
after Topchiev (IPCS)

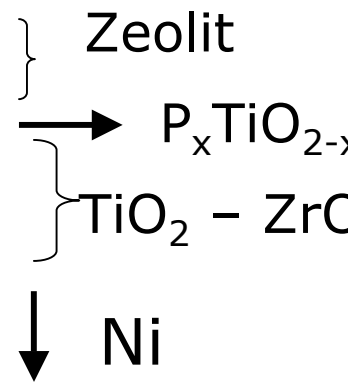
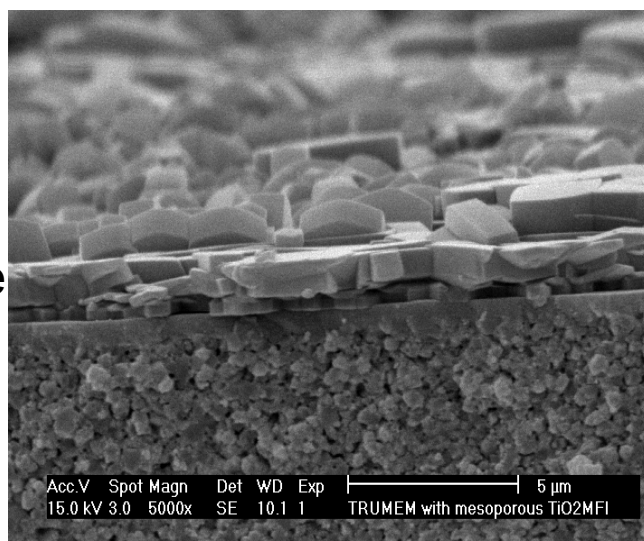
Institute of Chemical Physics
Research (ICPR)

Institute of Problems of Micro-
electronics Technology and High
Purity Materials (IPM)

Moscow Power Engineering Institute
(MEI)

Academy for Fine Chemistry
Technologies (MINCT)

- ❖ CMR Fuel Cells (UK)
- ❖ Lawrence Livermore National Laboratory (USA)
- ❖ National Physical Laboratory (UK)
- ❖ Fraunhofer Institute (Germany)
- ❖ Waterloo University (Canada)



Electrochemical Generator “Cascade-IP”

Independent Power Technologies, Ltd.



The generator is based on advanced Alkaline Fuel Cell modules. One of the unique features of the generator is a patented zero-waste regenerative scrubber used for the removal of carbon dioxide from the incoming air. The scrubber reduces the operational cost and increases the overall serviceability of the generator.

Performance

Maximum power: 6.0 kW; Voltage: 30.0 – 42.5 V

Maximum current: 200 A; Fuel: Hydrogen

H₂ consumption < 4.5 m³/h; Air consumption < 36 m³/h

Pressure: Atmospheric

Operating conditions

Ambient temperature: -20 to +40 °C

Relative humidity (max): 98% at 25 °C

General

Dimensions: 1250 x 582 x 863 mm; Gross weight: 240 kg

CO₂ scrubber: built-in, regenerative

Start-up: self-starting with built-in hydrogen burner < 15 min at 20 °C

Electrolyte: 6.6 M KOH (aqueous solution)



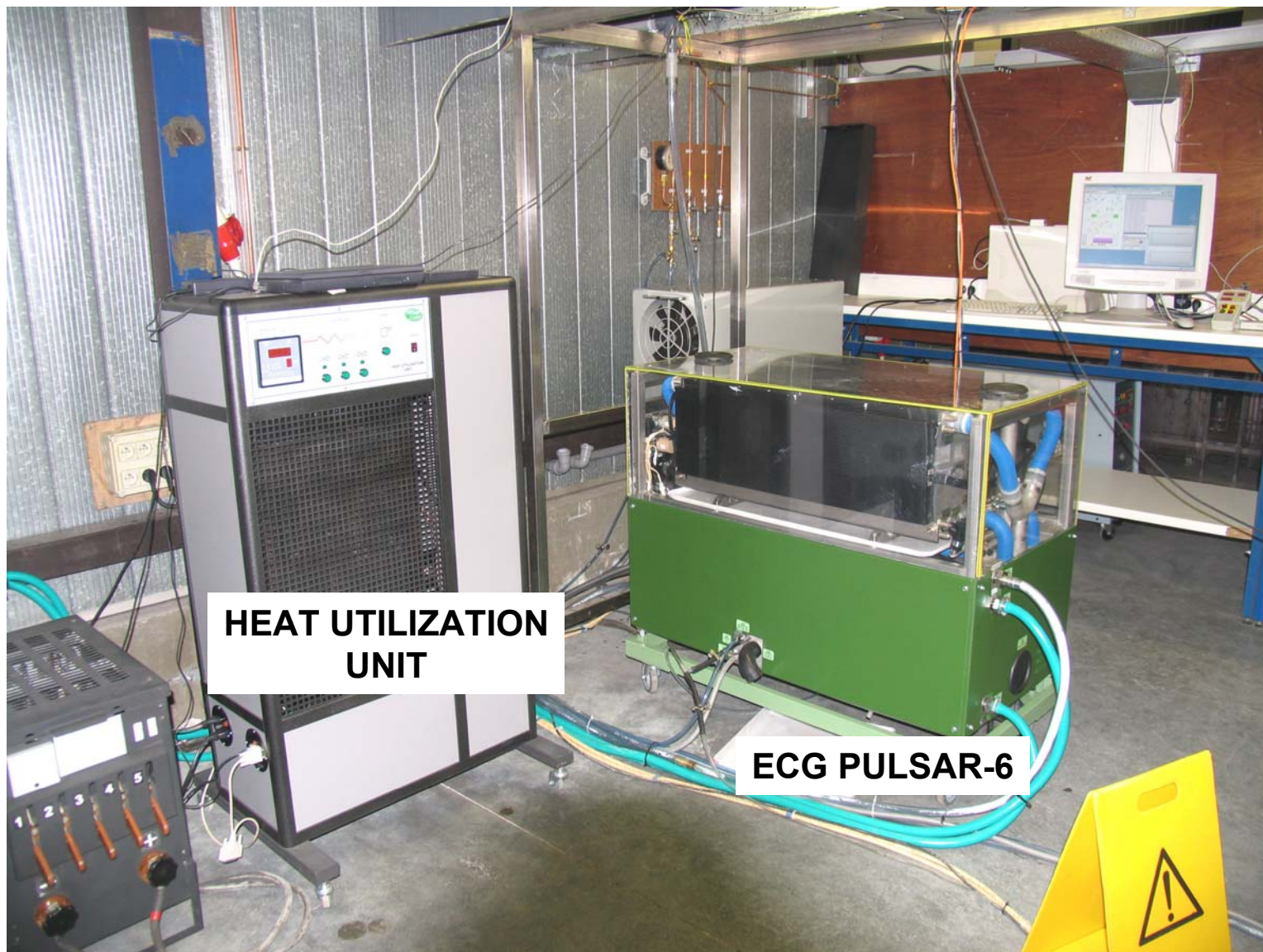
The scrubber

Independent Power Technologies Ltd

Address: 3-d Mytishchinskaya 16, bldg 60 129626, Moscow, Russia

Tel: (7-095)2312109; Fax: (7-095)2312078; E-mail: karichev@independentpower.biz

PULSAR-6 CO- GENERATOR



**HEAT UTILIZATION
UNIT**

ECG PULSAR-6

Solid Oxide Fuel Cell



Nominal Power 2,5 kW

Voltage 43,2 V

**Reached Power 2 kW
at 36 V**

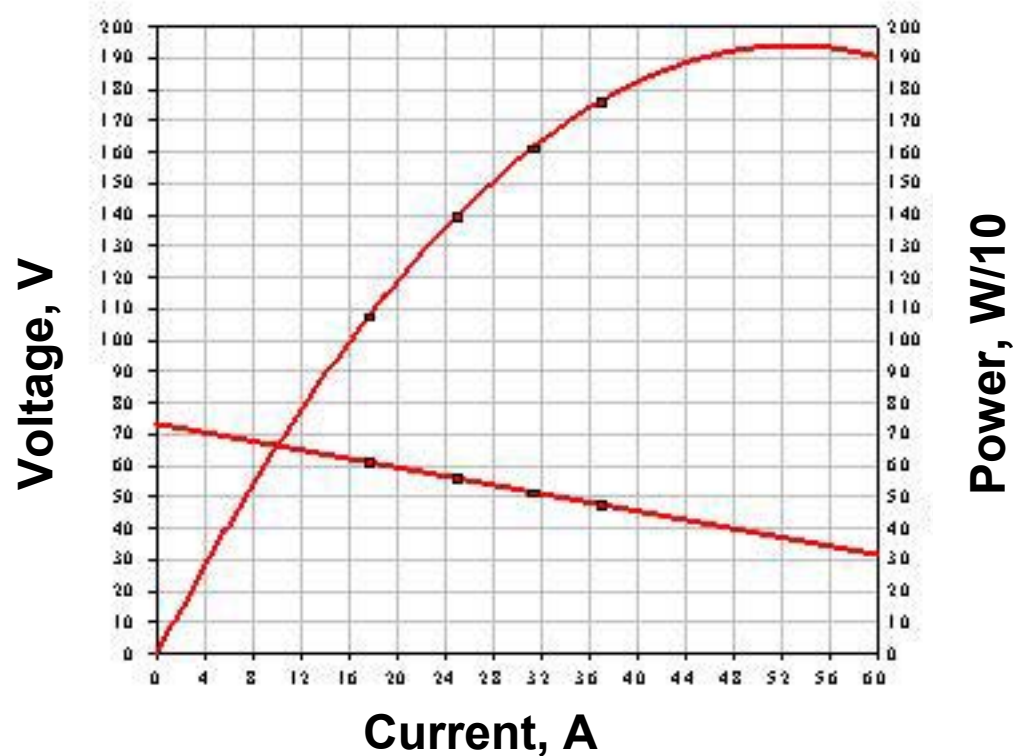
P.O. 245, Snezhinsk, Chelyabinsk reg.
456770 Russian Federation
+7(351-46)54367 fax: +7(351-46)55566
www.vniitf.ru

Solid Oxide Fuel Cell

Power unit



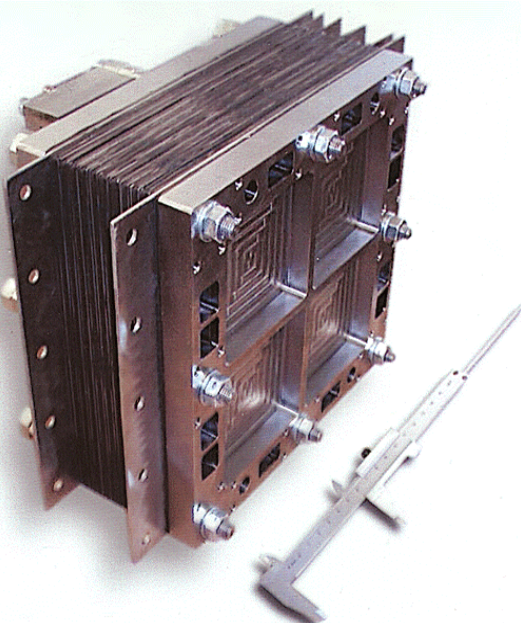
Test results



PEM hydrogen fuel cells

Hydrogen – Air fuel
cell.

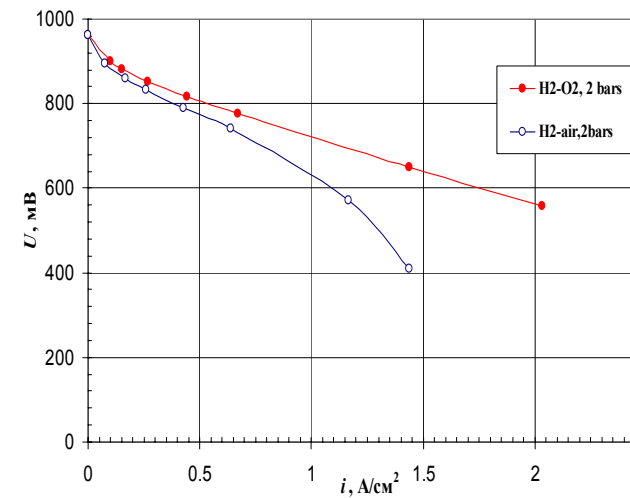
Power 1kW



Hydrogen – Air fuel
cell.

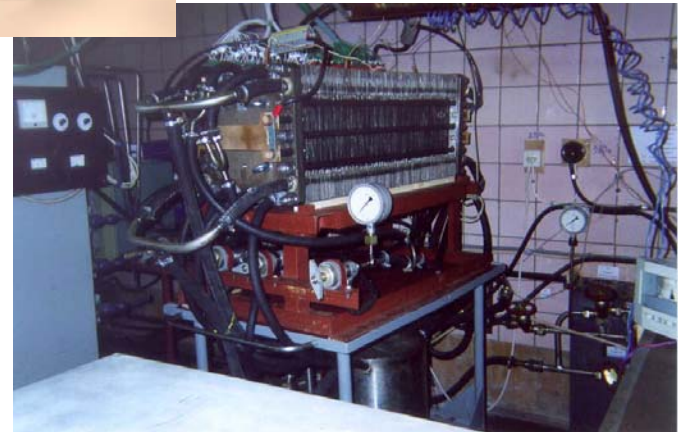
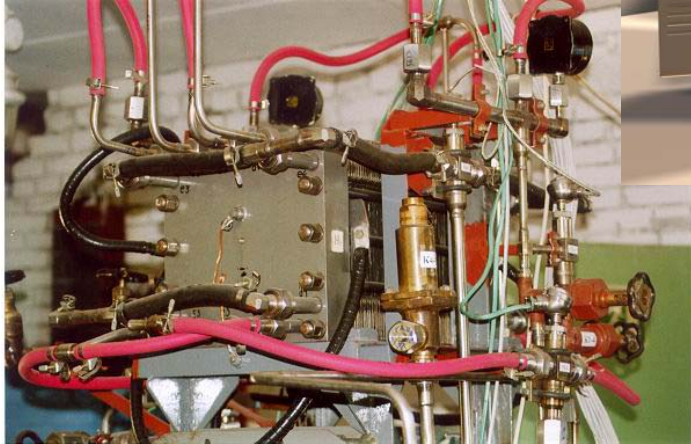
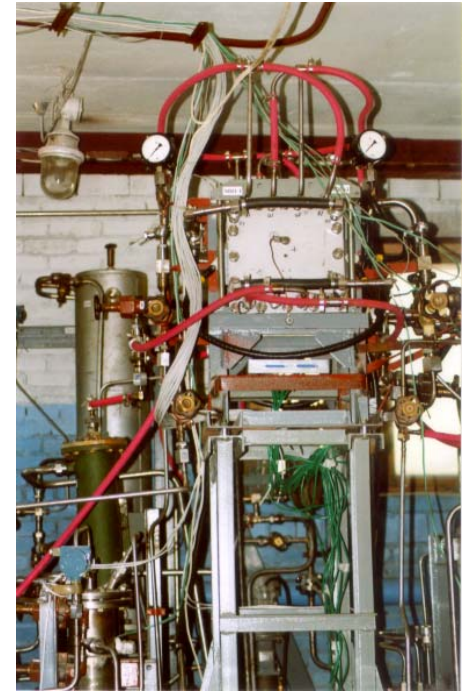
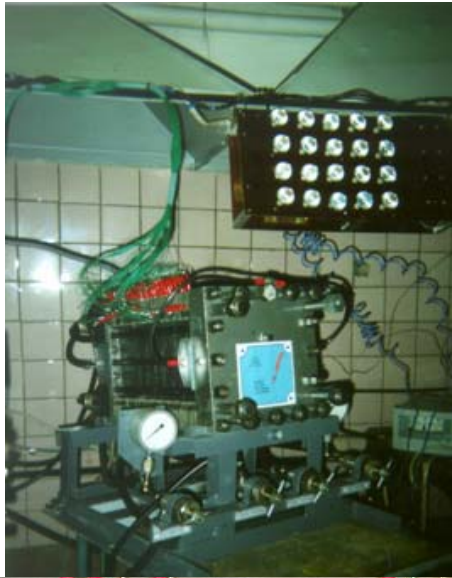
Power 0.2 kW

Polarization curves for
PEMFC (Typical MEA
performance)



PEM hydrogen fuel cells

Central Research Institute for
Ship Electric Engineering and
Technology, Saint-Petersburg



Hydrogen Storage



FASI Program in Hydrogen Storage

Program: Research and Innovations in Priority Trends of Scientific and Technical Development for 2002-2006

Priority direction: Energy and Energy Saving

Project: **R&D of new technologies of safe solid-state hydrogen storage on the base of reversible metal hydrides and nanostructured composites**

Participants:

Joint Institute for High Temperatures RAS,
Institute for Problems of Chemical Physics RAS,
Ural Institute of Metal Physics RAS,
Moscow State University,
Moscow Power Engineering Institute.

IPHE International Project Development of reversible solid state hydrogen storage system for fuel cell power supply system



Joint Institute for High Temperatures RAS

IPHE Contact Person: Dr. Stanislav Malyshenko

Project Coordinator: Vasily Borzenko,

Joint Institute for High temperatures RAS (IVTAN)

Krasnokazarmennaya 17a, Moscow, 111116 Russia

litp@dataforce.net www.litp.ru

Dr. Dmitry Dunikov, Joint Institute for High Temperatures, RAS (Russia)

Dr. Boris Tarasov, Institute for Problems of Chemical Physics RAS (Russia),

Prof. Anatoly Yermakov, Ural Institute of Metal Physics RAS (Russia),

Dr. Sergey Mitrokhin, Moscow State University (Russia),

Dr. Georgy Yankov, Moscow Power Engineering Institute (Russia),

Prof. John R. Lloyd, Michigan State University (US),

Prof. James Wang, Sandia Nat'l Labs (US),

Prof. Peixue Jiang, Tsinghua University (China),

Prof. Volodymyr A. Yartys, Institute for Energy Technology (Norway),

Prof. Thorsteinn I. Sigfusson, University Iceland (Iceland),

Prof. Hirohisa Uchida, Tokai University (Japan),

Prof. Allan Schroeder Pedersen, Risoe National Laboratory (Denmark),

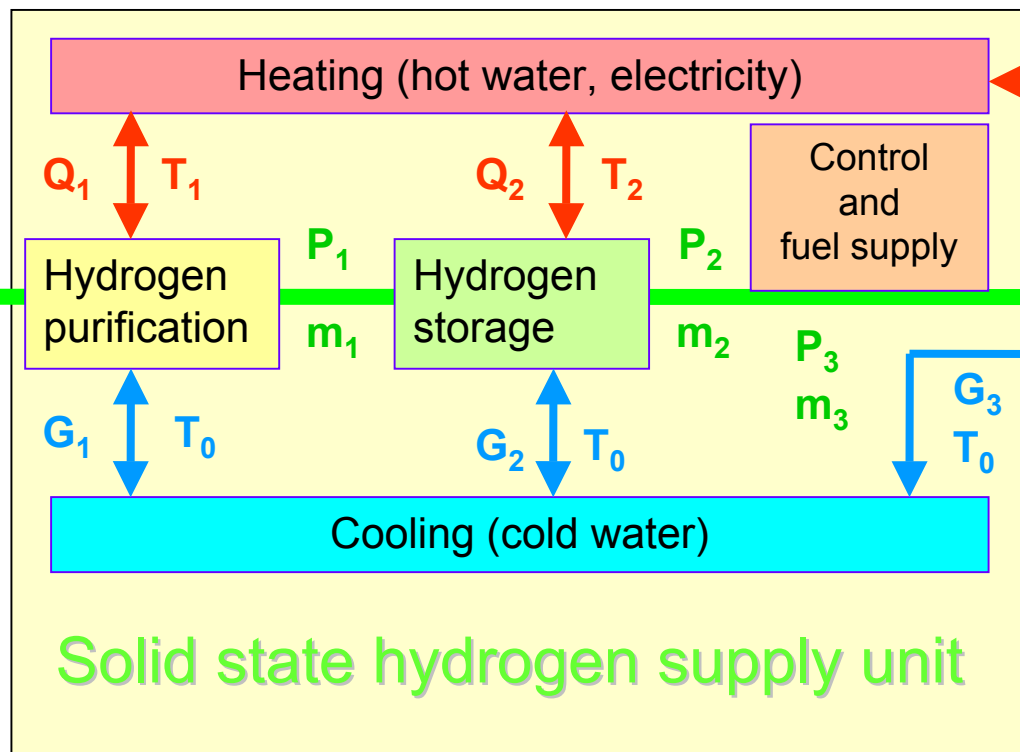
Prof. Kee Suk Nahm, Chonbuk National University (Rep. of Korea).



Development of reversible solid state hydrogen storage system for fuel cell power supply system



Primary hydrogen supply



Fuel Cell

Fuel demand $m(t)$

Power demand $E(t)$

AC/DC

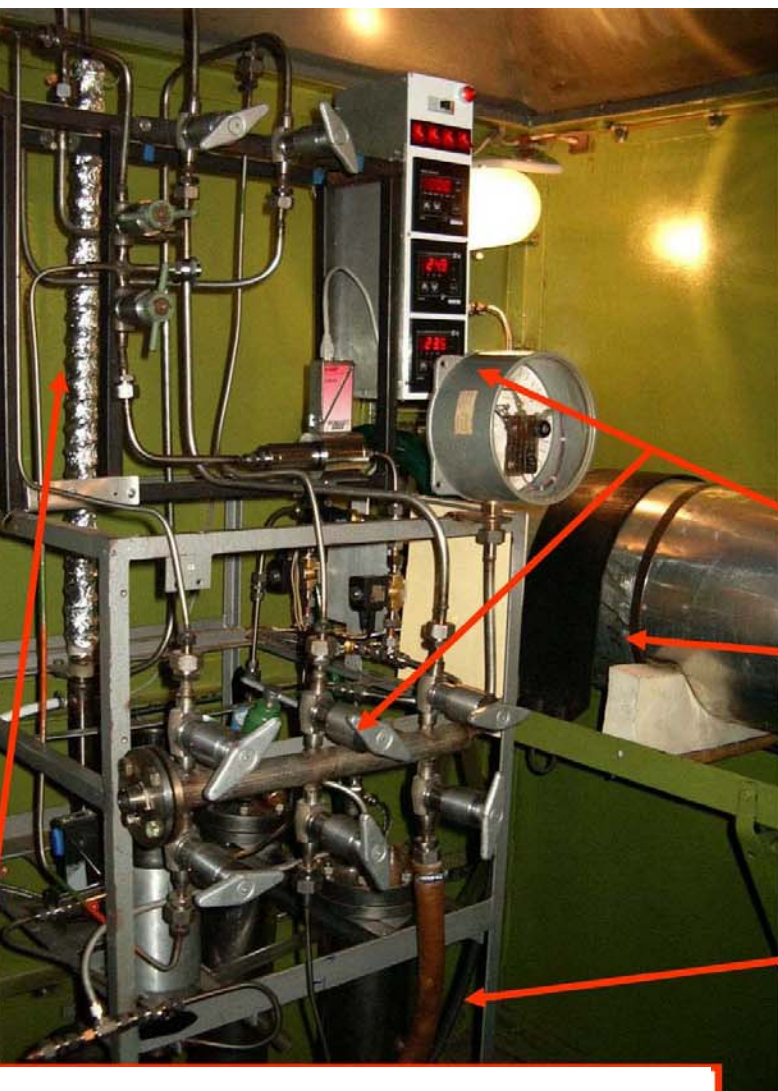
Consumer

FC type	T_3	T_1, T_2	T_0
High temperature	< 500 °C	> 100 °C	~ 20 °C
Low temperature	< 100 °C	~ 80 °C	~ 20 °C

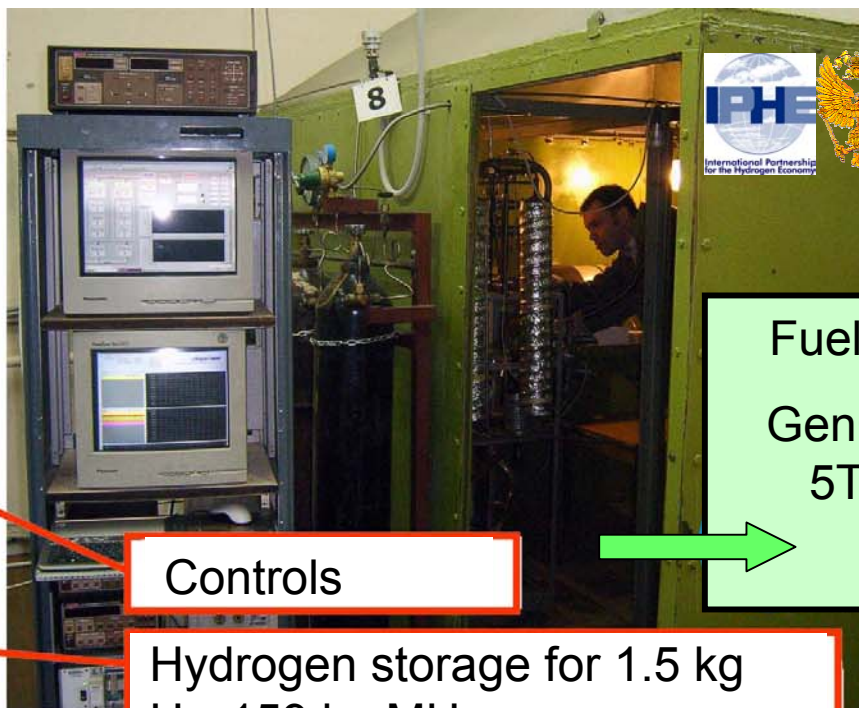
Concept of FC power supply unit with solid state hydrogen storage



VTTAN H2Lab experimental facility for complex investigations of solid-state reversible hydrogen storage and purification systems



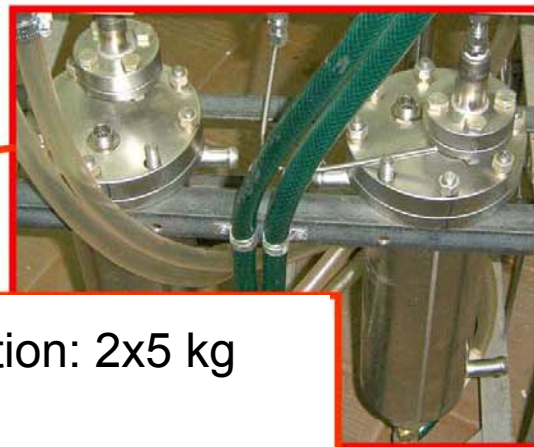
Preliminary purification: catalytic burner and dryers



Controls

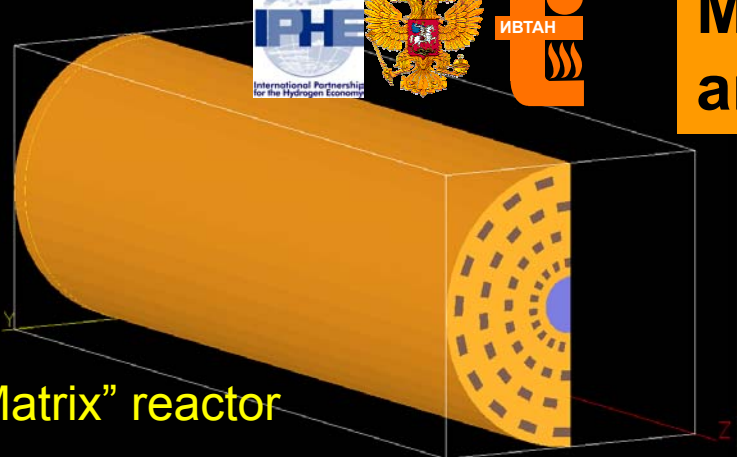
Hydrogen storage for 1.5 kg H_2 , 159 kg MH

Fuel Cell
GenCore
5T48

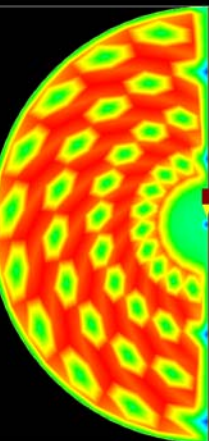


Fine purification: 2x5 kg
MH reactors

cylindrical reactor
 ed with MH powder
 (yellow), water
 cooling $T_0 = 20\text{ }^\circ\text{C}$ –
 bes and outer
 perimeter, hydrogen
 et – inner porous
 be (blue)



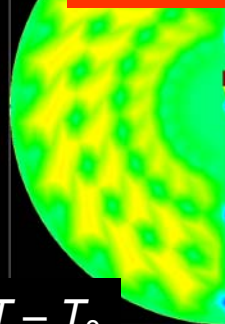
“Matrix” reactor



$t = 10\text{ s}$

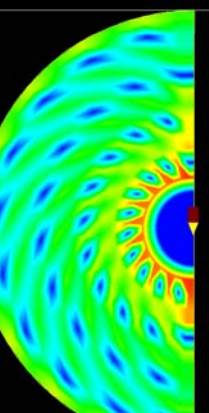
6.68
 10
 13.4
 16.7
 20
 23.4
 26.7
 30.1
 33.4
 36.8
 40.1
 43.4
 46.8

$t = 300\text{ s}$



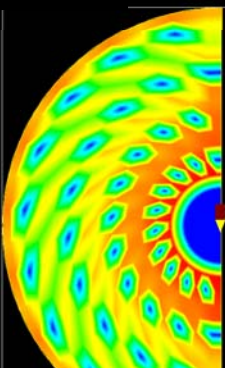
Temperature fields $T - T_0$

Stored hydrogen [H/Me]



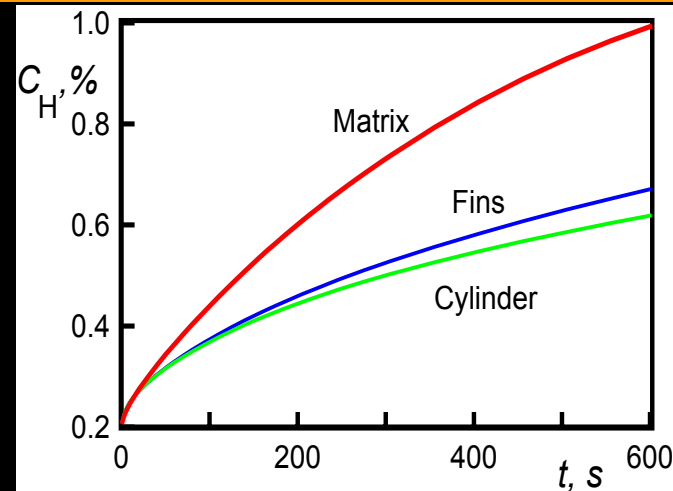
$t = 300\text{ s}$

value:
 0.843
 1.12
 1.39
 1.67
 1.94
 2.22
 2.49
 2.77
 3.04
 3.31
 3.59
 3.86
 4.14
 4.41
 4.69
 4.96

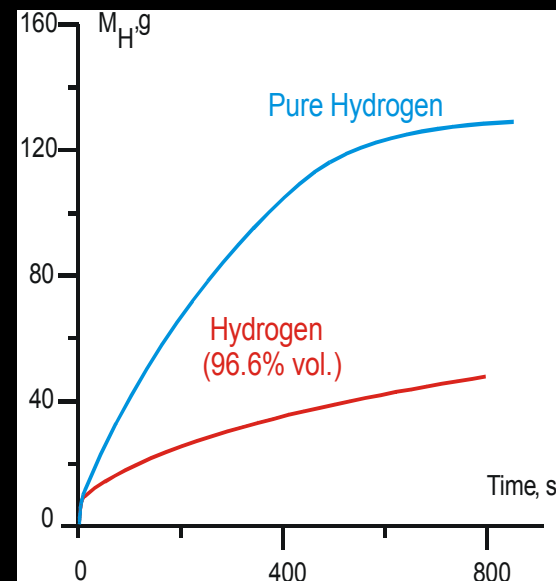


$t = 600\text{ s}$

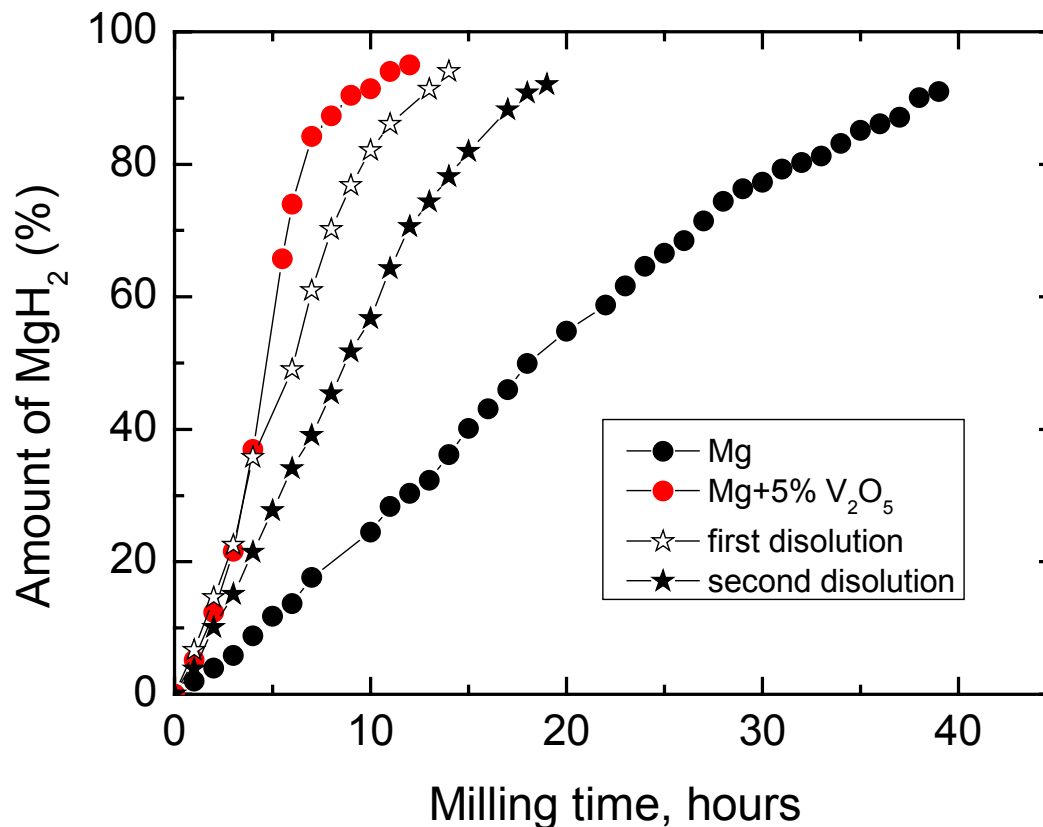
Mathematical modelling and reactor optimization



Effect of heat transfer enhancement



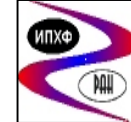
Presence of impurities dramatically decreases efficiency



Hydrogen uptake at MA of pure Mg
with different content of a catalyst under H_2 pressure 15 bar

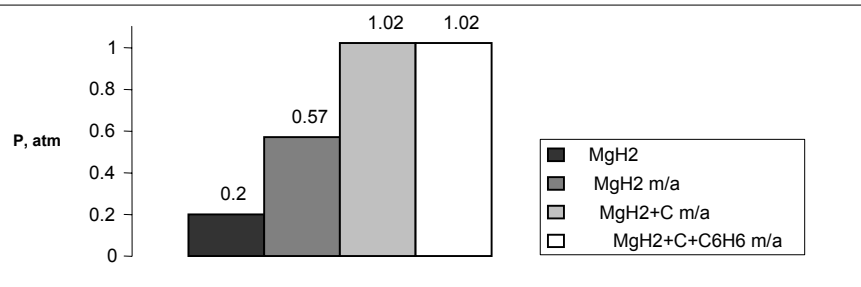
Hydrogen Storage Materials

Institute for Problems of Chemical Physics RAS

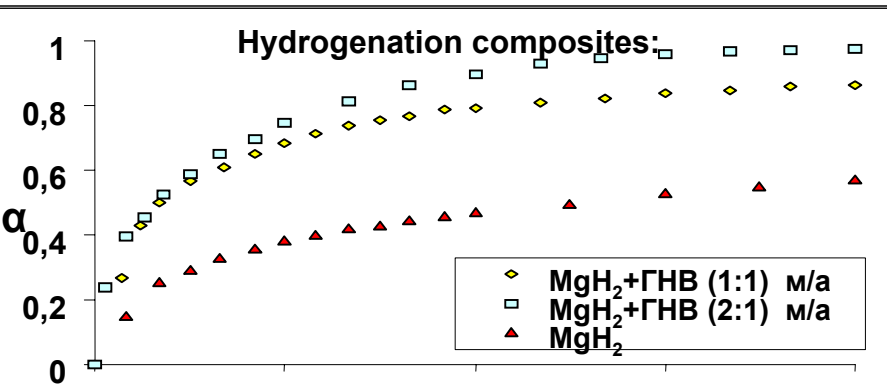
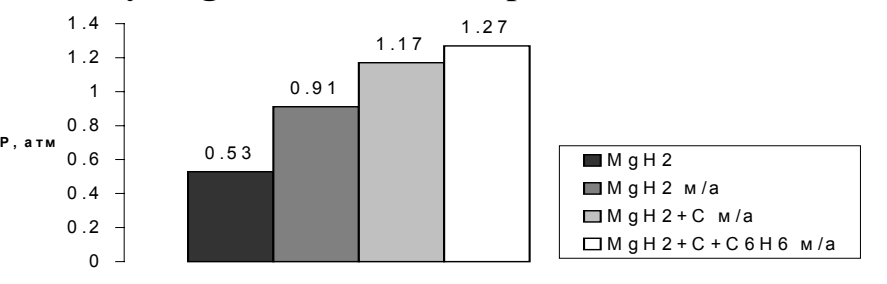


Composites based on magnesium hydride and carbon

Dehydrogenation at the temperature of 150°C:



Dehydrogenation at the temperature of 250°C:



- Volume – 600 cm³
- Loading density MH – 60% (1.26 g/cm³)
- Weight MH – 500 g
- Calculated hydrogen capacity – 280 l

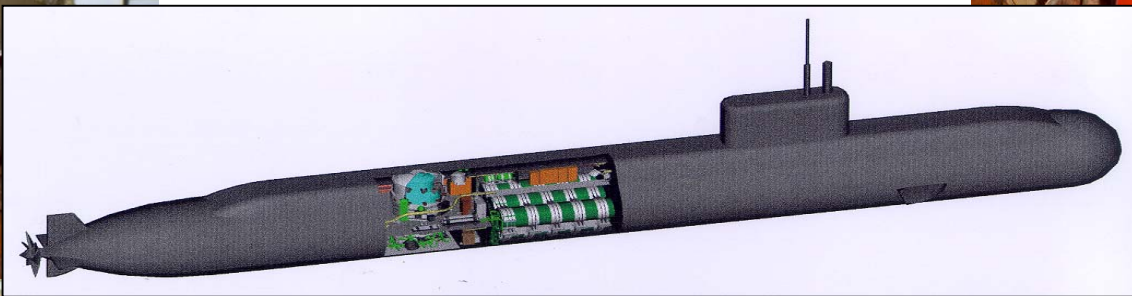
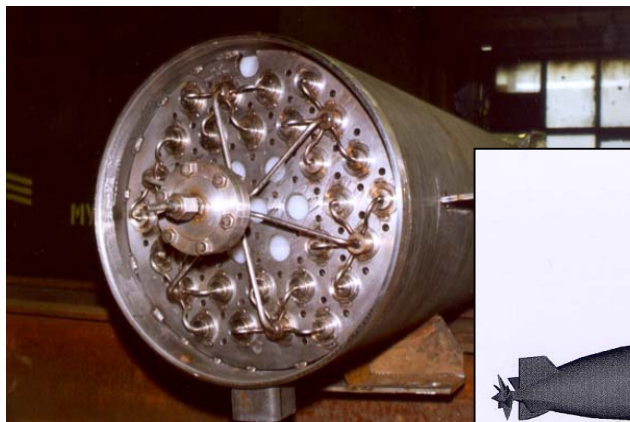
Characteristics of a hydrogen accumulator based on magnesium alloy



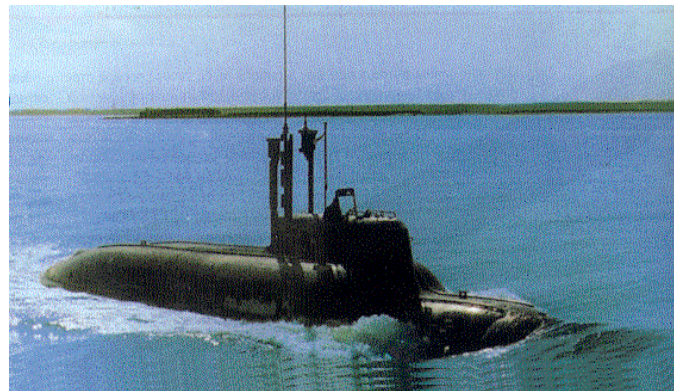
- Hydrogen sorption – 250 °C / 10 bar
- Hydrogen desorption – 350 °C / 0-2 bar
- Duration of 90% charge - < 10 min.
- Duration of 90% discharge - 30 min.
- Reversible capacity – 200 l
- Total capacity – 280 l

Hydrogen Storage

Central Research Institute for
Ship Electric Engineering and
Technology, Saint-Petersburg

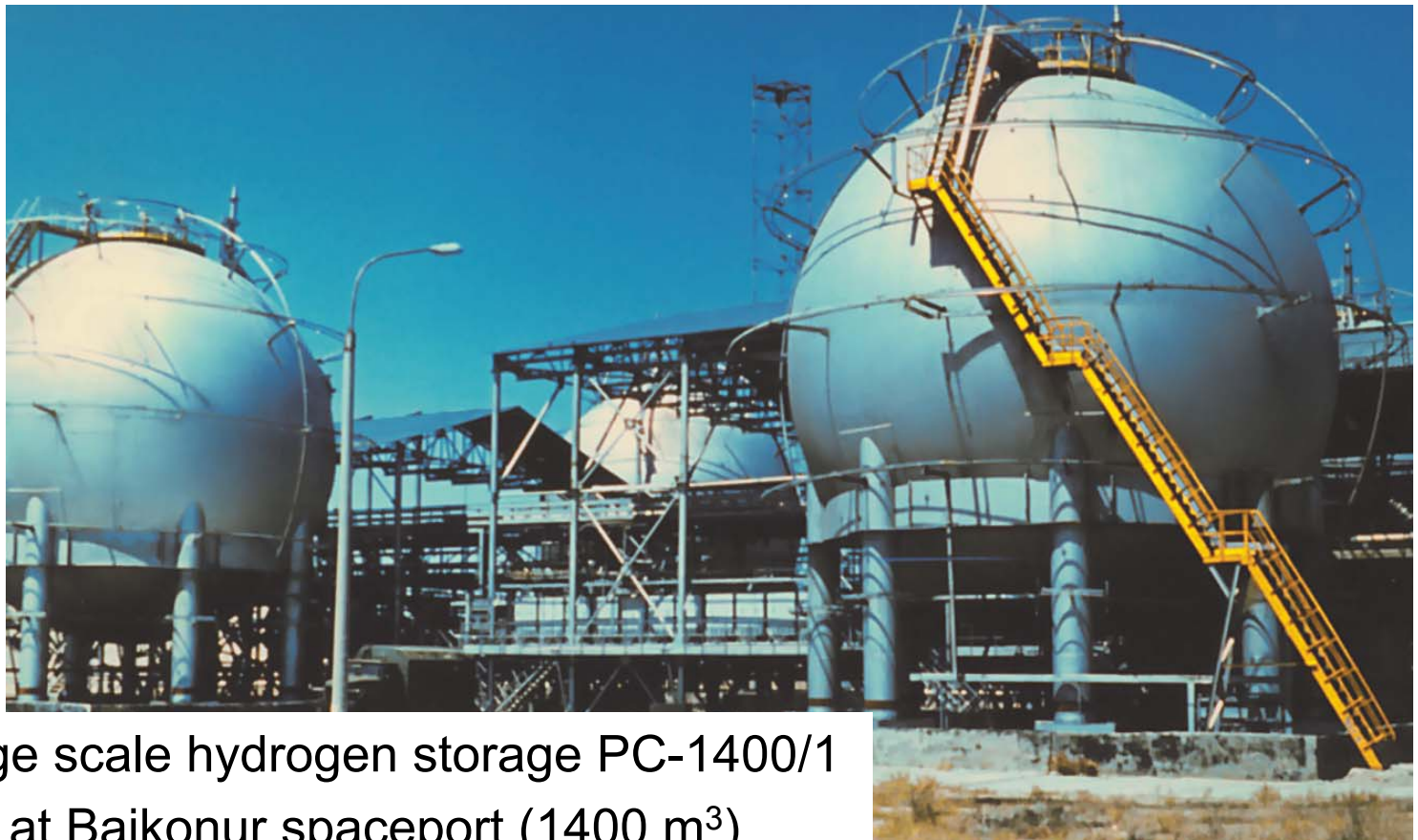


Hydrogen storage systems for
more than 1,000 kg of
intermetallic alloy



CRYOGENIC HYDROGEN STORAGE

JCR Cryogenmash



Large scale hydrogen storage PC-1400/1
at Baikonur spaceport (1400 m³)

143907, Russia, Moscow region, Balashikha, Lenin av. 67
+7 495 521-5722

e-mail: root@cryogenmash.ru www.cryogenmash.ru

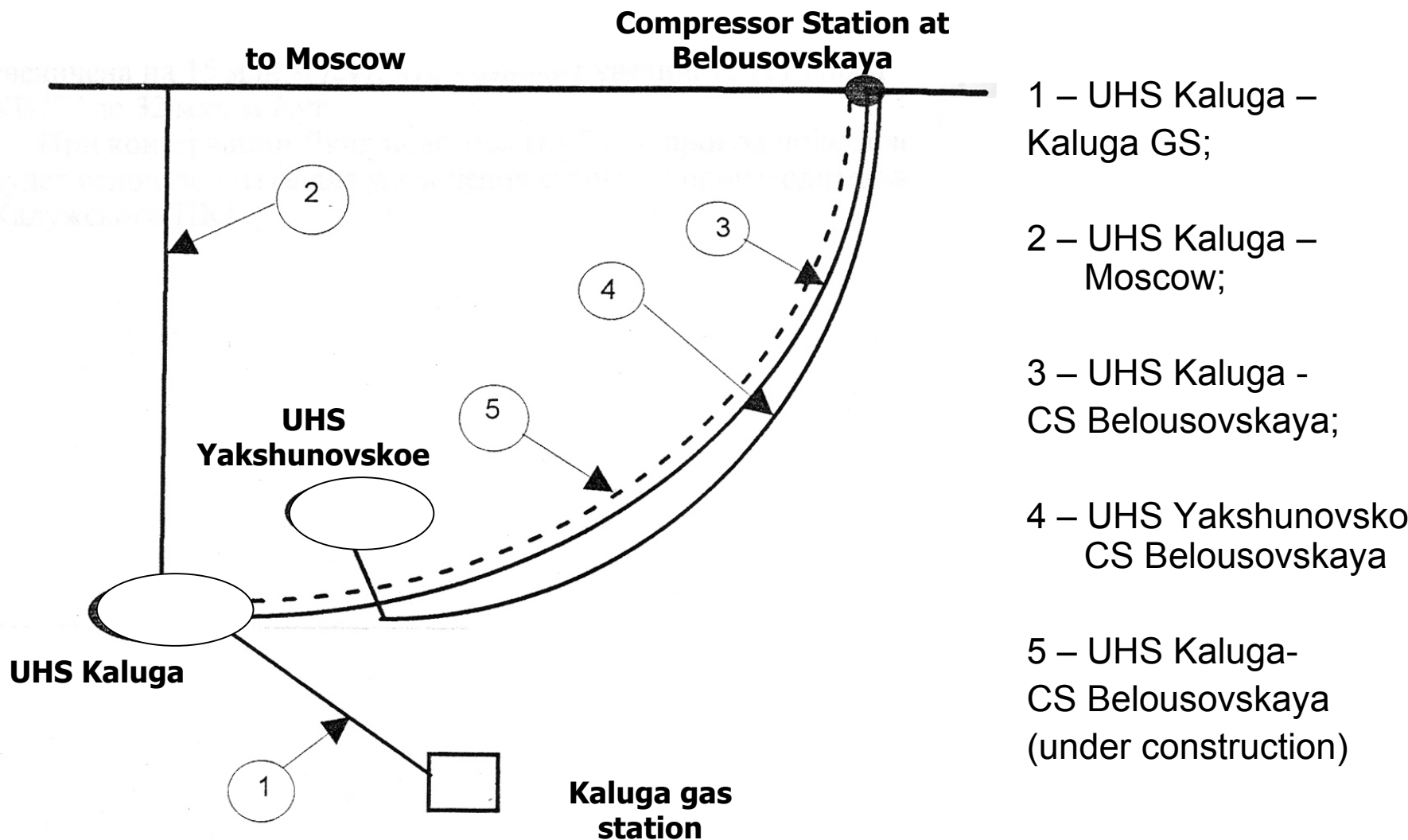
Cryogenic hydrogen tanks characteristics

JSC “Cryogenmash”

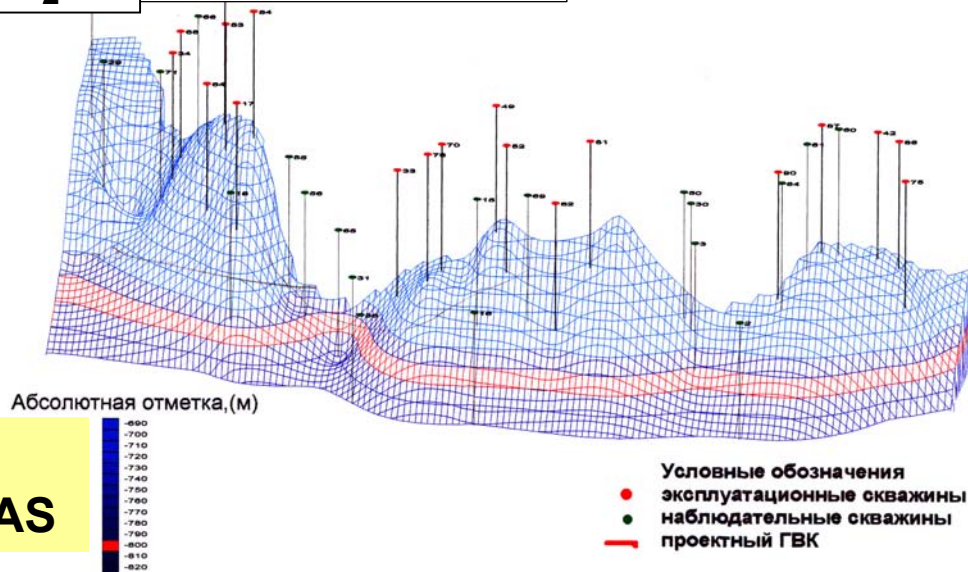
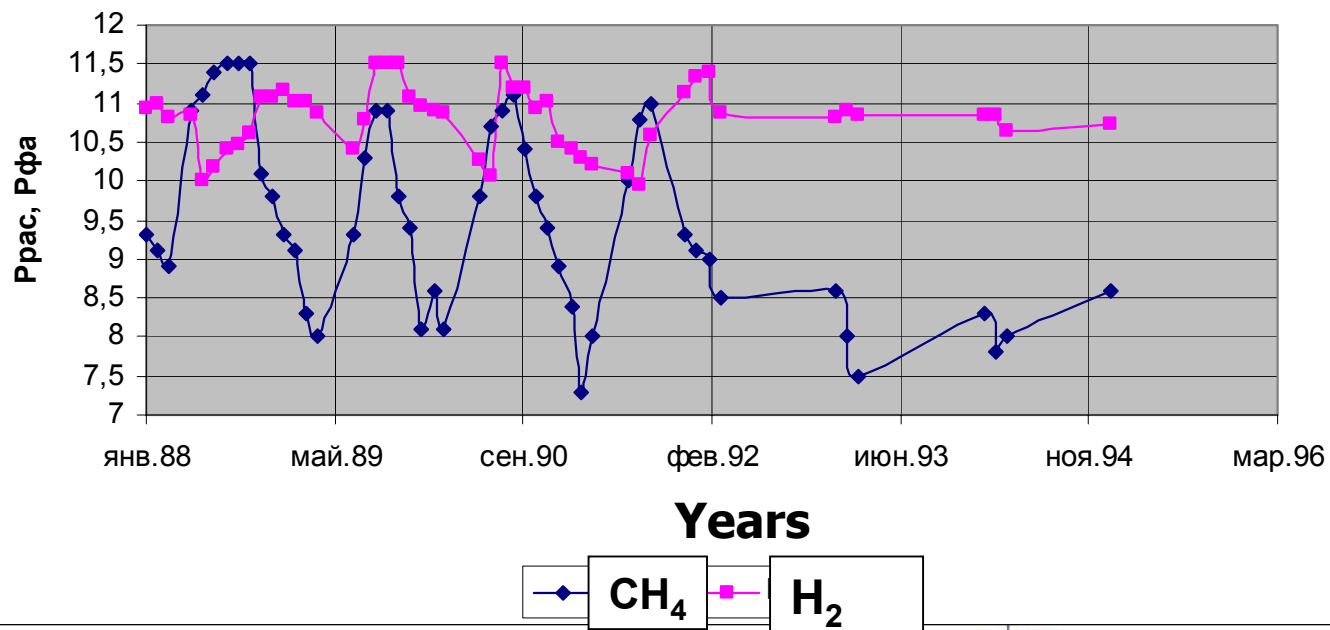
Parameter	РЦБ-63/0.5	РЦГ-250/1	РЦГ-250/0.8	РСБ-1400/1	ЦТВ-25/0.6 т	ЦТВ-45/1.0 т
Capacity, m ³	66.3	246	246	1,437	25	45
Working pressure,MPa	0.5	1.0	0.85	1.05	0.7	1.0
Product mass,t	4.43	15.7	15.7	96.79	1.5	2.74
Evaporation loss daily, %	0.52	0.3	0.35	0.13	1.2	1.0
Dimensions, m						
length		36.35	36.35	-	14.35	15.7
width	3.68	3.7	37.4	-	2.75	2.5
height	12.65	3.92	3.92	20.1	3.92	3.7
Tank mass, t	22	72	72	360	19	21.76

143907, Russia, Moscow region, Balashikha, Lenin av. 67
+7 495 521-5722
e-mail: root@cryogenmash.ru ww.cryogenmash.ru

Underground Hydrogen Storage (UHS) in Kaluga Region



Load/Reload of H₂ и CH₄ on UHS “Yakshunovskoe”

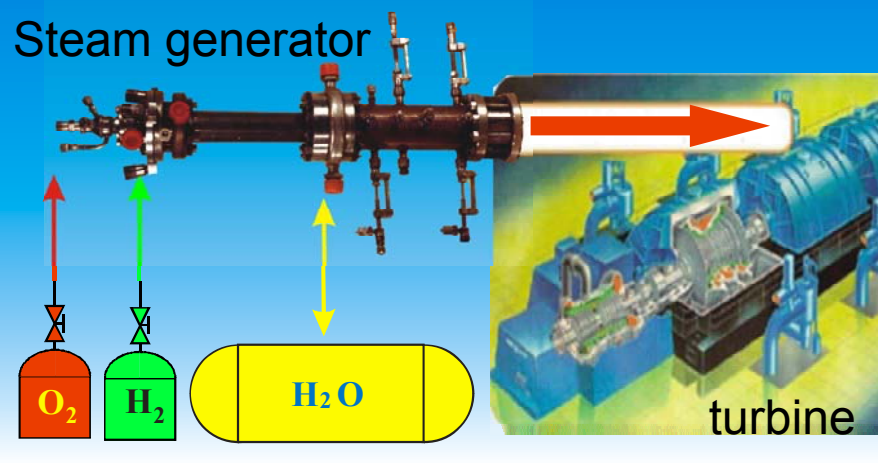
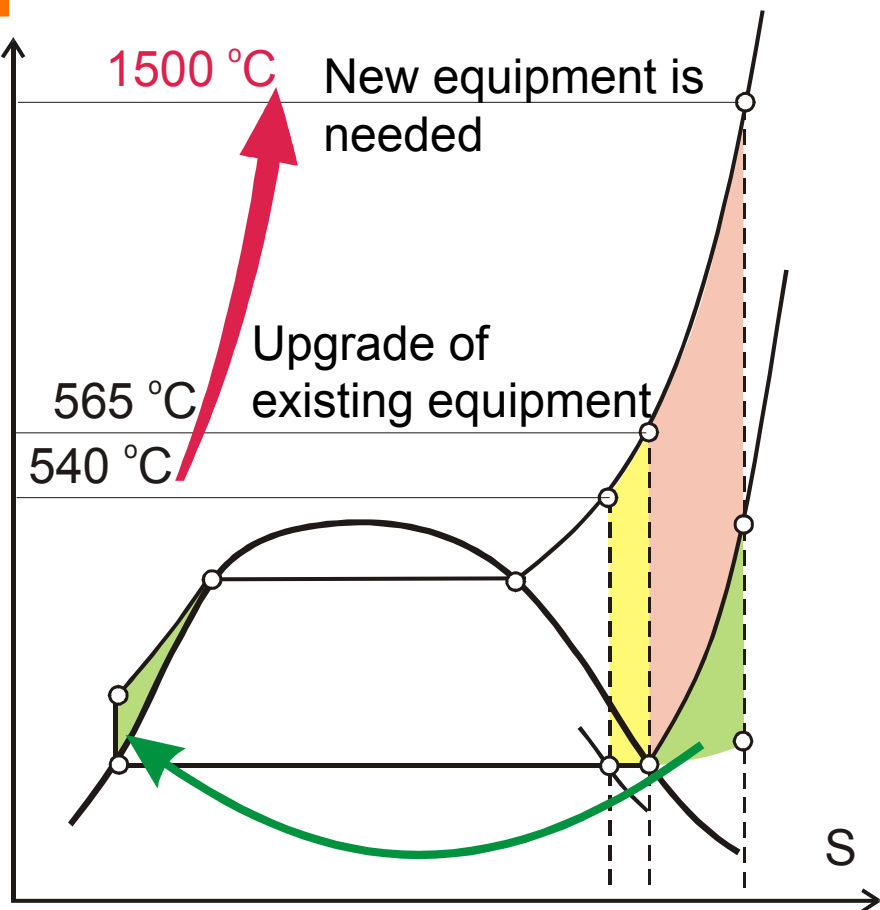


Hydrogen Combustion Systems

Hydrogen combustion systems



Institute for High Temperatures RAS



Upgrade of existing equipment with hydrogen-oxygen steam generators will help to increase of overall power plant efficiency with high efficiency of hydrogen utilization

Development of new high temperature steam turbines and heat regeneration systems will result in great increase of overall power plant efficiency with high efficiency of hydrogen utilization

Key problem: to create high temperature and high pressure hydrogen-oxygen generators working on the stoichiometry component mixture

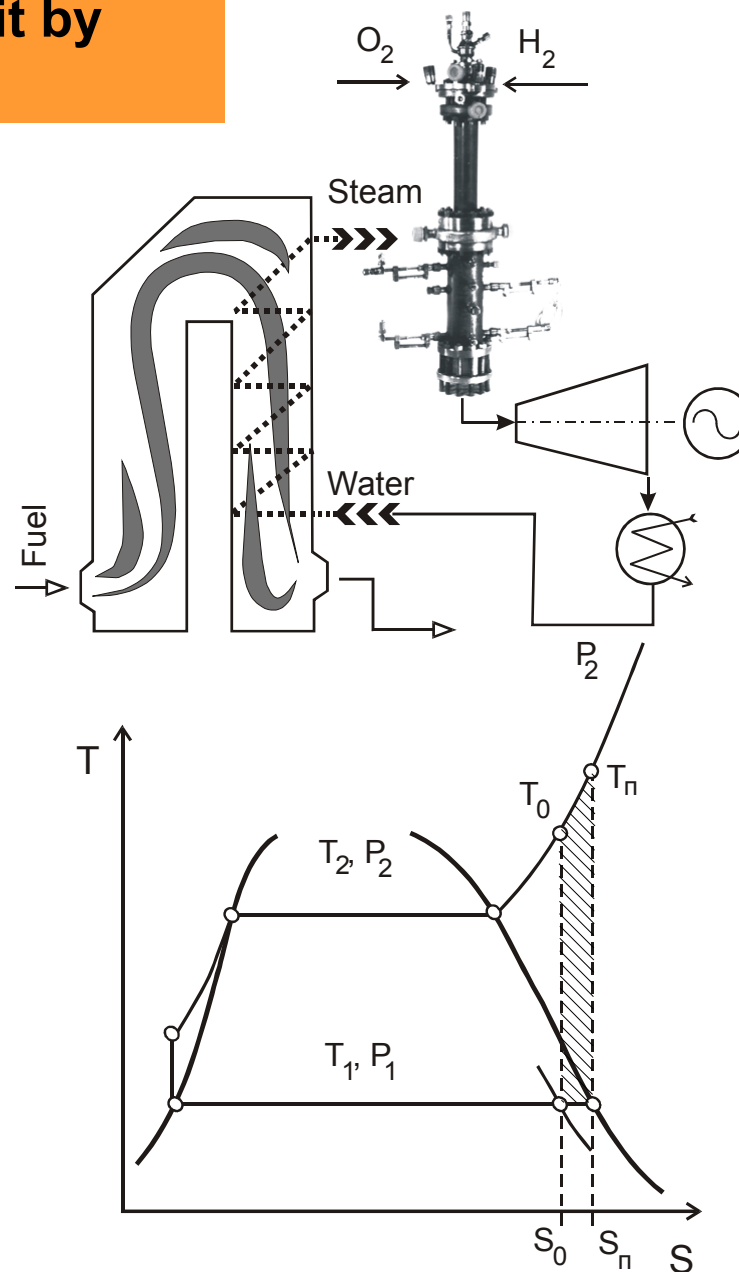
Upgrade of K-200-130 steam turbine power unit by hydrogen superheating of steam

Power plant:

Steam production – 640 t/h, power – 200 MW,
 $P = 130 \text{ atm}$, $T = 540^\circ\text{C}$,
 Fuel consumption – 275 g.c.e./kW h

Superheating from 540°C to 565°C

H_2/O_2 gas consumption, $10^3 \text{ st.m}^3/\text{h}$	11.5/5.82
Additional power ΔW , MW	20.1
Fuel consumption, g.c.e./kW h	262
Overall efficiency without hydrogen superheating, %	40.3
Overall efficiency with hydrogen superheating, %	42.3
Efficiency of hydrogen utilization, %	57.5



Experimental high-pressure H₂/O₂-steam generators



Parameters of experimental fire blocks

Model	Thermal power	Steam parameters	
		<i>T</i> , K	<i>P</i> , MPa
20 K	20-100 kW	1100	0,5
100 K	100-150 kW	1000	4
10 M	10-20 MW	1200	7
25 M	25-30 MW	1200	10

Small-scale high-pressure H₂/O₂-steam generators for use in stationary and mobile autonomous hydrogen zero emission power units (ZEPU) based on 30-150 kW mini-turbine.

The main features of experimental device:

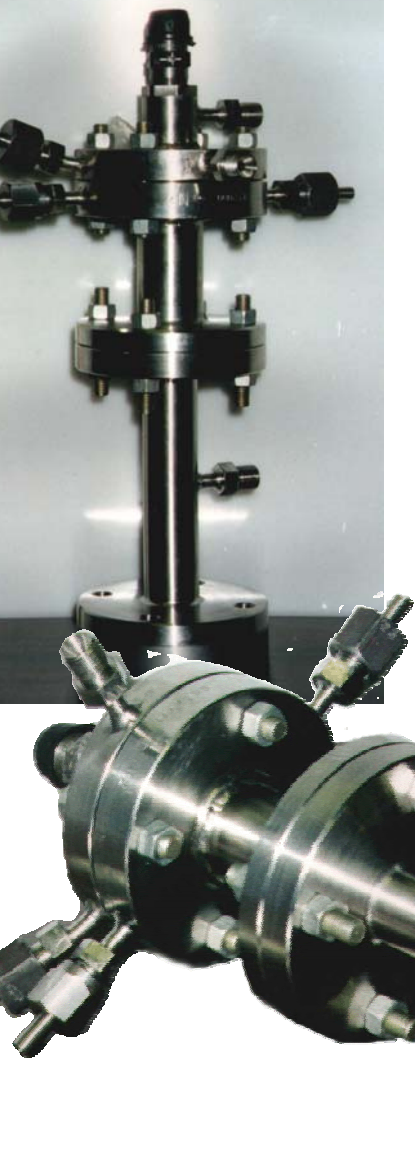
working pressure, MPa: 1-4

steam temperature, K: 600-1000

thermal capacity, kW: 40-156

length, mm: 300

max diameter, mm: 90.



**Joint Institute for High Temperatures RA
Chemical Automatics Design Bureau
Keldysh Center**

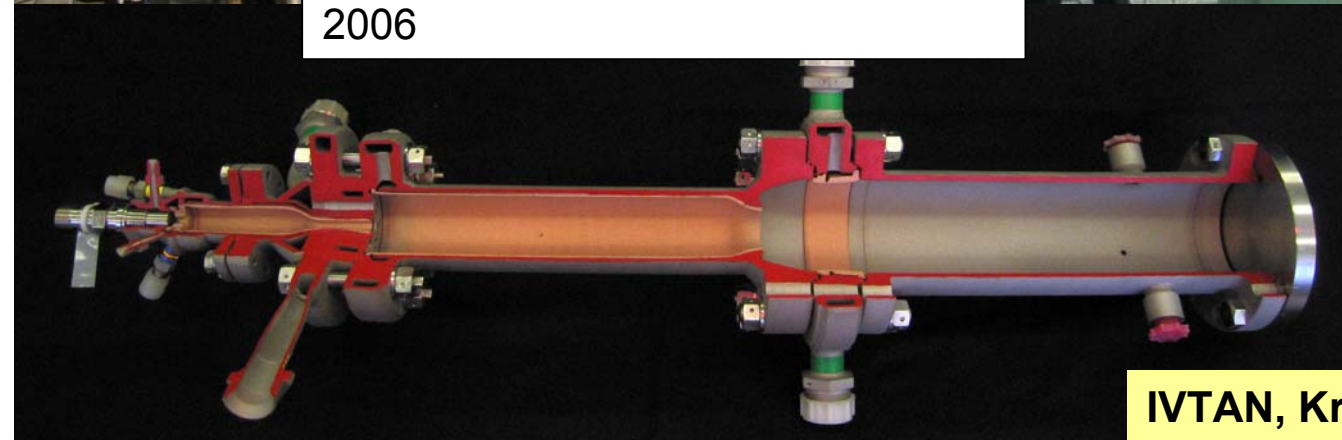
Experimental MW class steam generators



Joint Institute for High Temperatures RAS



Fire tests on Chemical Automatics
Design Bureau site, Voronezh,
2006



IVTAN, Krasnokazarmennaya 17a
Moscow 111116 Russia
litn@dataforce.net

Liquid hydrogen
Production facility.
View of construction
site.



Firing tests of rocket engines
at tilted test stand

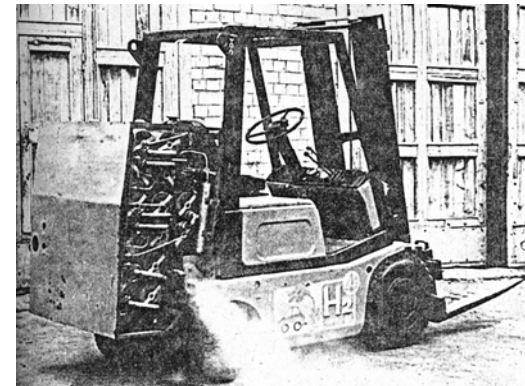
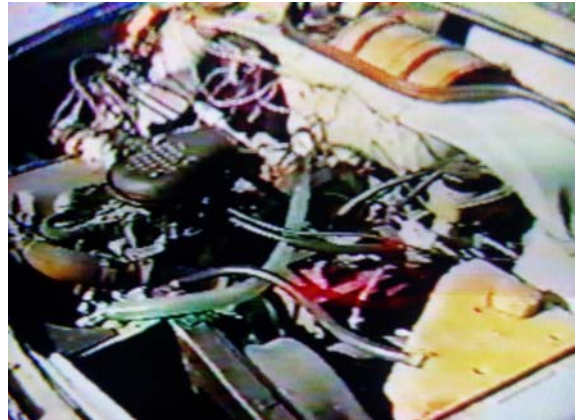
Contact: Chemical Automatics Design Bureau, Russian Space Agency

Address: Voroshilova st. 22, Voronezh, 394006 Russia

Tel.: (7-0732) 333673 Fax: (7-0732) 334122 E-mail: cadb@comch.ru

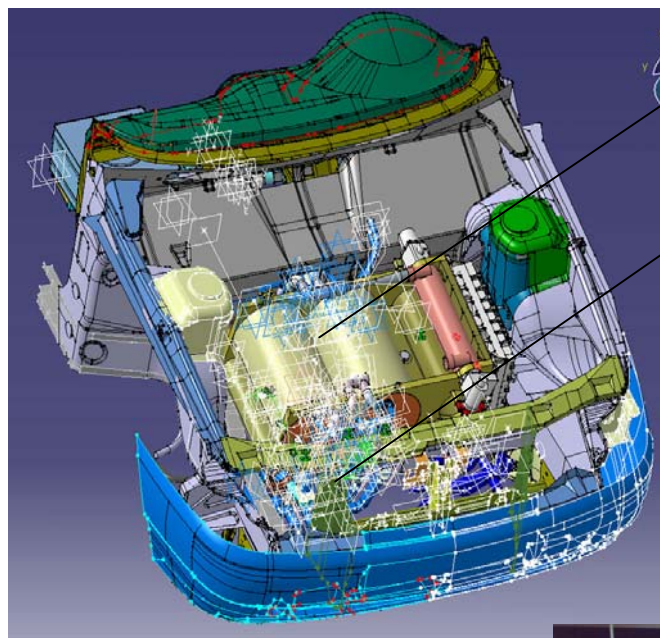
Hydrogen for Vehicles

Internal Combustion vehicles



Fuel Cell Vehicles

RCC Energia, UEChC, Autovaz

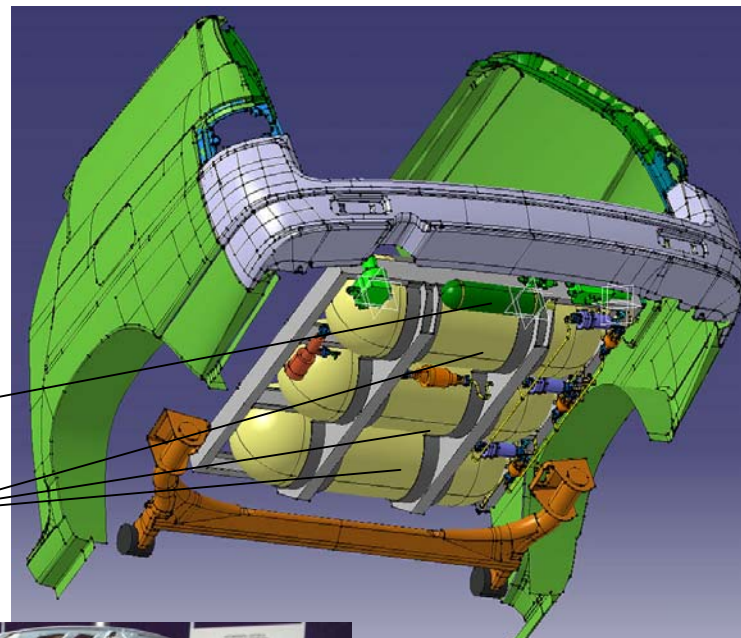


Alcaline FC

Radiator

Compressed
Nitrogen

Compressed
Hydrogen



MAIN CHARACTERISTICS

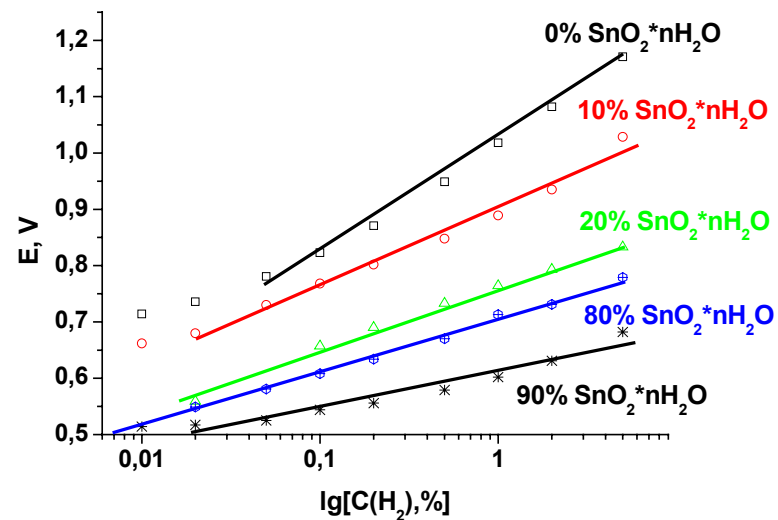
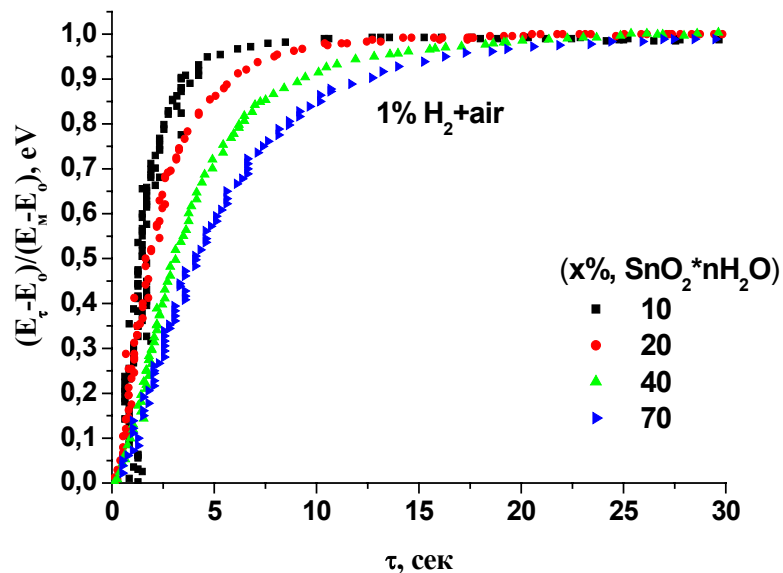
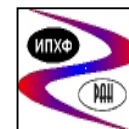
	«ANTEL-1»	«ANTEL-2»
FUEL	HYDROGEN+ OXYGEN	HYDROGEN+ AIR
MECHANICAL POWER, kW	15	18
MAXIMUM POWER, kW	25	40
VOLTAGE, V	115-150	180-200
HYDROGEN PRESSURE, ATM	300	400
RANGE, KM	220	300



SAFETY AND STANDARDS

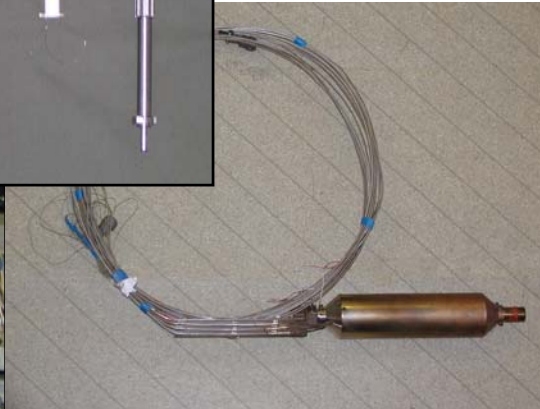
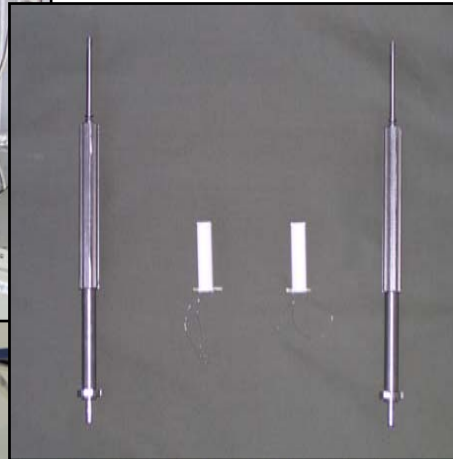
Electrochemical hydrogen sensors

Institute for Problems of Chemical Physics RAS



Hydrogen and Oxygen Gas Sensors

SSC "Institute for Physics & Power Engineering"



249033, Bondarenko sq. 1, Obninsk,
Kaluga reg., Russia

Combustion, detonation and explosion of hydrogen

Institute for High Energy Densities,
Associated Institute for High Temperatures RAS
125412, Moscow, Russia, Izhorskaya 13/19



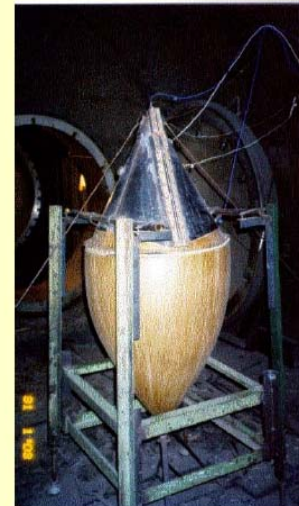
Explosive chamber



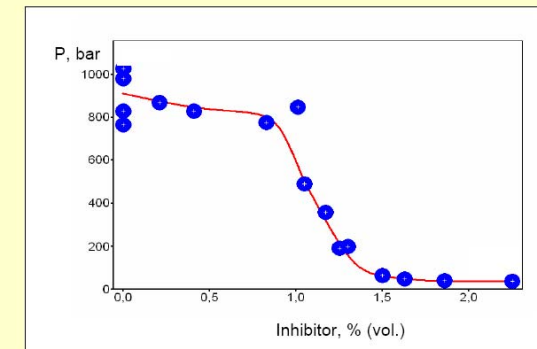
Inside diameter – 12 m; Wall thickness – 100 mm; Weight – 470 t.
Chamber is designed for explosion up to 1000 kg TNT.



Experimental installations “Cone” and “Pyramid”

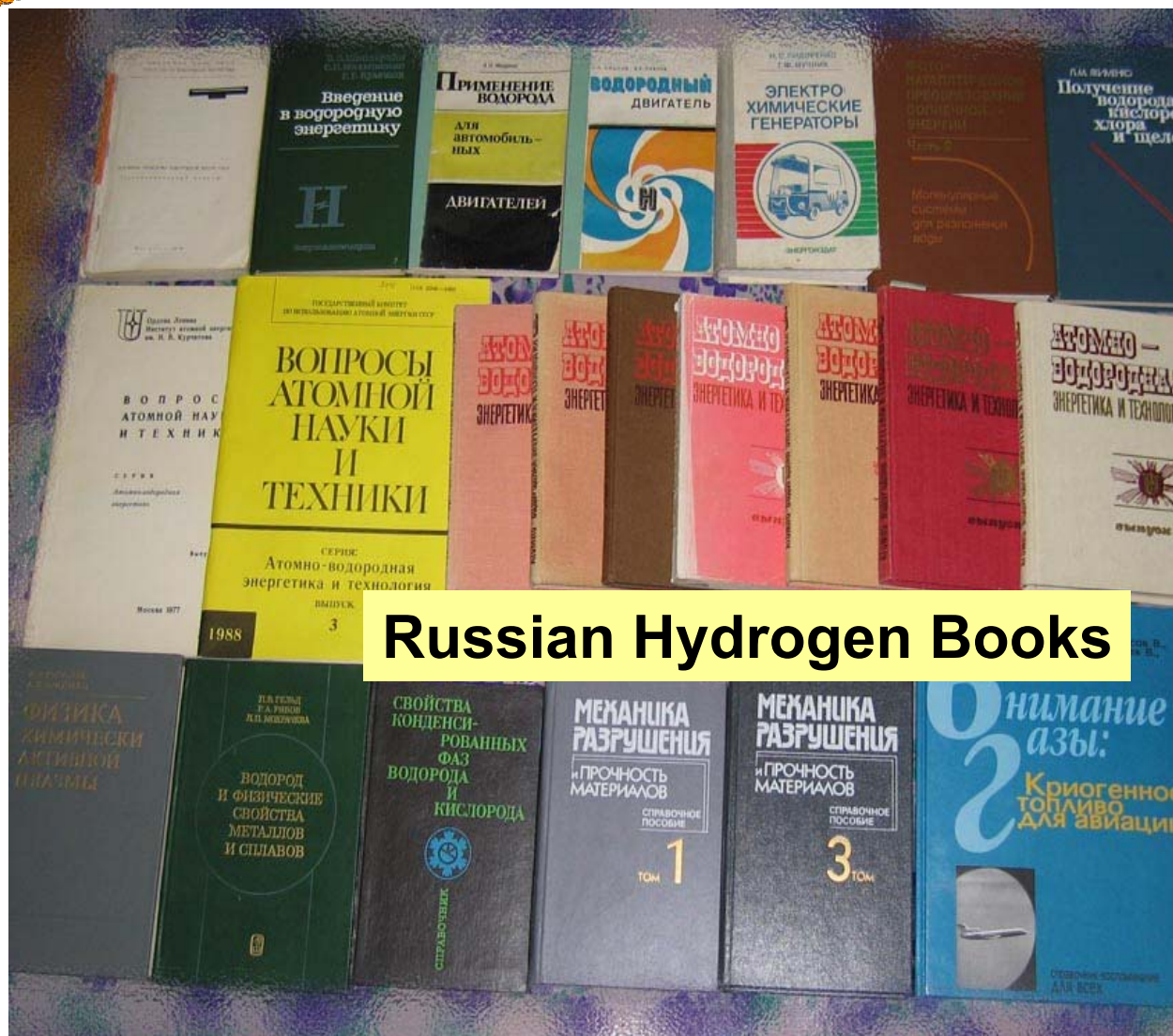


Suppression of explosion by Inhibitor





Education



Russian Hydrogen Books

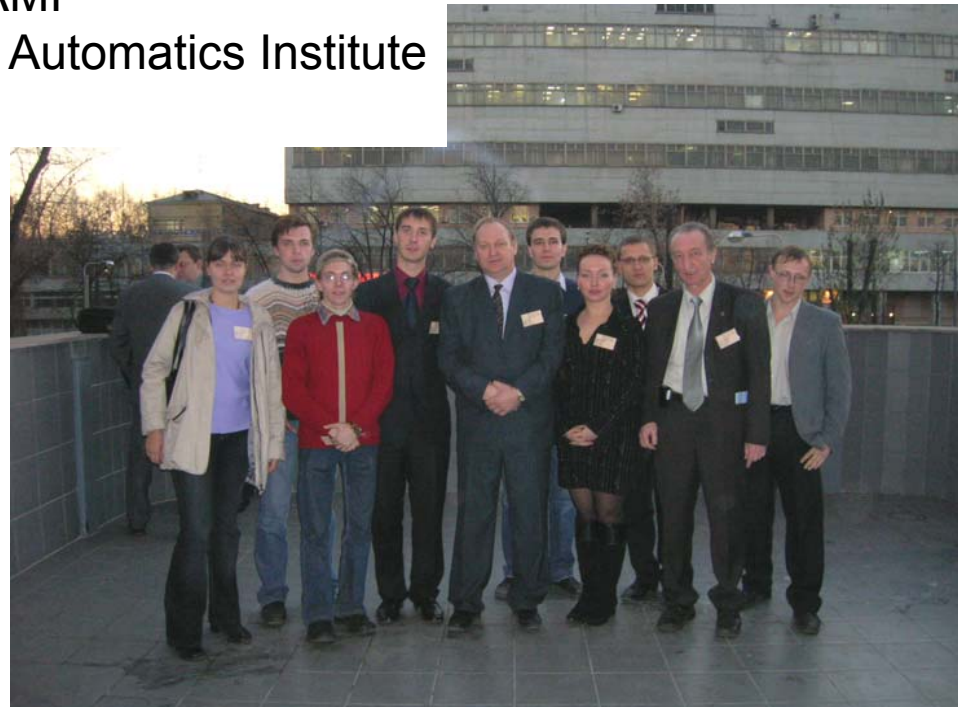


Hydrogen education in high school

FASI Project "Education"

Moscow Power Engineering Institute
Moscow Institute of Physics and Technology
Moscow State University
Saint-Petersburg State University
Novosibirsk State University
Ural Technical University (Ekaterinburg)
Tomsk Technical University
Moscow State Technical University "MAMI"
Moscow Radiotechnics Electronics and Automatics Institute
and many others...

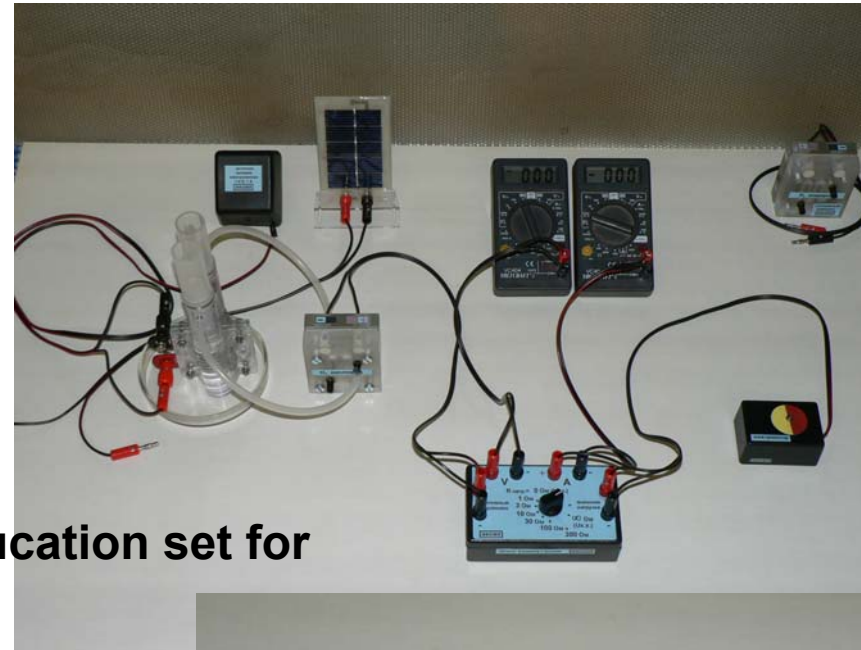
Lectures
Hydrogen Labs
Young Scientists Conferences
Students Hydrogen Clubs



Hydrogen education in elementary school



Hydrogen experimental and education set for elementary schools



Electrolyzer
Photovoltaic cell
Fuel Cell
Diagnostics



Results of Russian R&D in 2005-2006 and new 2007-2012 Federal Program are the good basis for EU-Russia cooperation in Hydrogen Energy

Thanks for your attention!